

# **CURRICULUM - 2023**

## **C -23**

### **DIPLOMA IN**

### **ELECTRICAL AND**

### **ELECTRONICS ENGINEERING**



**STATE BOARD OF TECHNICAL EDUCATION & TRAINING**  
**ANDHRA PRADESH**

## PREAMBLE

Technical Education is a key driver of economic development and plays a crucial role in providing individuals with the skills and knowledge necessary to thrive in the workplace. As technological advancements continue to reshape industries and create new opportunities, it is critical that technical education curricula remain relevant and up-to-date.

The curriculum has been designed with this in mind, with a focus on practical skills, critical thinking, and problem-solving. We believe that these skills are essential for success in both academic and professional spheres. The revamping of the technical education curriculum is made with collaborative effort from educators, industry experts, policymakers, and students.

At the heart of the curriculum, is the belief that the technical education should be **student-centered**, empowering learners to take ownership of their learning and pursue their passions. We aim to create a learning environment that is safe, supportive, and nurturing, where every student has the opportunity to reach their fullest potential. We acknowledge that learning is a lifelong journey, and our curriculum is designed to provide a solid foundation for continued growth and development. We hope that our students will not only leave with a diploma but with employability and passion for learning.

The State Board of Technical Education and Training, (SBTET) AP, has been offering Diploma programmes to meet the above said aspirations of the stake holders: industries, students, academia, parents and the society at large. **The Curriculum should be flexible, adaptable, and responsive to the changing needs of the industry and society.** As such, it has been the practice of SBTET, A.P., to keep the curriculum abreast with the advances in technology through systematic and scientific analysis of current curriculum and bring out an updated revised version at regular intervals.

The design of Curriculum C-23 was started in the month of January - 2023. Feedback was collected from all stake holders: Students, Lecturers, Senior Lecturers, Head of Sections and Principals for all programmes for this purpose. Accordingly, a workshop was convened on 15<sup>th</sup> February 2023 by Smt. C. Naga Rani, I.A.S, Director of Technical Education & Chairperson, SBTET, AP to discuss on revamping of C-20 curriculum to meet the needs of industries and for improvement of placements.

The meeting was attended by Sri. Saurab Gaur, I.A.S, Principal Secretary, Skill Development & Training, Smt. Lavanya Veni, I.A.S, Director, Employment & Training. Thirteen Representatives from Industries and Fourteen Academicians from Higher Level Institutions and officials of ITI, Skill Development, CTE & SBTET attended the workshop.

Smt. C Naga Rani, I.A.S., Commissioner of Technical Education while addressing in the workshop, emphasized the necessity of industrial training and on-hand experience, that the students need to undergo to support the industries and the Gaps in the Curriculum need to be fixed to make the students passionate to work in the industry in order to support economy of the country.

The committees of each branch consisting of experts from Industries, Higher Level Institutions and Faculty of Polytechnics are informed to study the possibility of incorporating the following aspects while preparation of the curriculum so as to improve employability.

- **To bring out industry oriented Diploma Engineers.**
- **Internet of Things (IoT) for all branches**
- **Theoretical & Practical subjects 50: 50 Ratio**
- **Industry 4.0 concepts.**
- **5G Technology.**
- **Critical Thinking (Quantitative Aptitude, Data Interpretation, Quantitative reasoning etc) to face the written tests conducted by the industries during placements.**

In continuation, series of workshops with subject experts followed in the subsequent weeks for thorough perusal for preparation of draft curriculum. Also, the suggestions received from representatives from various industries, academic experts from higher level institutions, subject experts from Polytechnics, have been recorded, validated for incorporation into the **Curriculum C-23**. Finally, the draft curriculum was sent to academicians of higher-level institutions, industrial experts for Vetting.

The design of new Curricula C-23 for different diploma programmes has thus been finalised with the active participation of the members of the faculty teaching in the Polytechnics of Andhra Pradesh, and duly reviewed by Expert Committee constituted of academicians and representatives from industries. Thus, the primary objective of the curriculum change is to produce employable diploma holders in the country by correlating the growing needs of the industries with relevant academic input.

The outcome-based approach as given by NBA guidelines has been followed throughout the design of this curriculum and designed to meet the requirements of NBA Accreditation, too.

**The Revised Curriculum i.e., Curriculum-2023 (C-23) is approved by 45<sup>th</sup> Academic Committee of SBTET, A.P for its implementation with effect from Academic Year 2023-24. Also, the SBTET, A.P under the aegis of the Department of Technical Education, Andhra Pradesh in it's 62<sup>nd</sup> Board Meeting held on 13-07-2023 (vide item no: 17) Approved to update the Polytechnic Curriculum C-23 with effect from the academic year 2023-2024 onwards after revamping the present C-20 curriculum, to meet the latest industrial technological developments including Industry 4.0 concepts.**

## **2. HIGHLIGHTS OF CURRICULUM C-23**

*The following Courses/ Topics are incorporated in this curriculum C-23 as per the suggestions received from Industrial Experts, Faculty of Higher Level Institutions and Polytechnics to improve the Employability Skills of the Polytechnic Students.*

1. Duration of course for regular Diploma is 3 years.
2. The Curriculum is prepared in Semester Pattern. However, First Year is maintained as Year-wise pattern.
3. 6 Months Industrial training has been introduced for 3 years Diploma Courses in VI semester.

4. Updated subjects/topics relevant to the industry are introduced in all courses at appropriate places.
5. The policy decisions taken at the State and Central level with regard to environmental science are implemented by including relevant topics in Chemistry. This is also in accordance with the Supreme Court guidelines issued in Sri Mehta's case.
6. Keeping in view the increased need of communication skills which is playing a major role in the success of Diploma Level students in the industries, emphasis is given for learning and acquiring listening, speaking, reading and writing skills in English. Further as emphasized in the meetings, Communication Skills lab and Life Skills lab are continuing for all the branches.
7. CAD specific to the branch has been given emphasis in the curriculum. Preparing drawings using CAD software has been given more importance.
8. Upon reviewing the existing C-20 curriculum, it is found that the theory content is found to have more weightage than the Practical content. In C-23 curriculum, more emphasis is given to the practical content in Laboratories and Workshops, thus strengthening the practical skills. The ratio of Theory & Practicals is 50:50.
9. With increased emphasis for the student to acquire Practical skills, the course content in all the subjects is thoroughly reviewed and structured as outcome based than the conventional procedure based.
10. Curriculum of Laboratory and Workshops have been thoroughly revised based on the suggestions received from the industry and faculty, for better utilization of the equipment available in the Polytechnics. The experiments /exercises that are chosen for the practical sessions are identified to confirm to the field requirements of industry.
11. The theory and practical subjects are restructured to find room for new theory and practical subjects to meet the present the industrial needs.
12. As electric vehicles are the key technology to decarbonise road transport, it is important to learn about EV Technology. Hence, to meet the need of present technology a new subject titled "ELECTRIC VEHICLE TECHNOLOGY" is introduced in V semester.
13. A new laboratory titled "HYBRID POWER SYSTEMS LABORATORY" is introduced in IV semester in which industrial visits are made compulsory to bridge the gap between classroom learning and real-world circumstances and to aware the latest trends in industries which facilitates the students for better understanding of power system concepts.
14. To make the students effective and efficient in all aspects, three periods per week are allotted in every year/semester for STUDENT CENTRIC ACTIVITY in which student will be trained for placements or make use of library or participate in sports & games/clean & green etc.

#### **SPECIFIC CHANGES INCORPORATED IN PRESENT CURRICULUM C-23**

- a) The number of theory subjects in each semester is limited to 05 only by restructuring the related subjects/topics and deleting repeated/higher order topics. Similarly, the relevant laboratories are restructured to find room for new laboratories.
- b) The duration of engineering drawing is made 03 periods by reducing the syllabus which is not necessary for Electrical & Electronics Engineering students.
- c) To boost the technical knowledge for better understanding of theory concepts the ratio of Theory & Practical is made 50:50 in this C-23 curriculum.

- d) The Electrical Engineering Drawing I & II are restructured and made into one single electrical drawing subject by deleting the topics which cover in theory subjects to find space for introducing new laboratories.
- e) A new laboratory EE-410 is introduced in IV semester titled with HYBRID POWER SYSTEM LABORATORY in which power systems practicals are introduced.
- f) Industrial visits play a key role for technical students which help to bridge the gap between classroom learning and real-world job circumstances. Keeping this in view, the industrial visits are made compulsory in EE-410, HYBRID POWER SYSTEM LABORATORY and proper weightage is given for industrial visits.
- g) A new theory subject titled ELECTRIC VEHICLE TECHNOLOGY, EE-502 is introduced in V semester in which EV technology and battery technology topics are introduced to meet the present industrial needs.
- h) MATLAB practicals are introduced in MATLAB PRACTICE LABORATORY, EE-506 in V semester in which simulation practicals are incorporated.
- i) SCADA practicals have been incorporated in PLC & SCADA laboratory, EE-507 in V semester to throw light on importance of SCADA in power system.

### 3. ACKNOWLEDGEMENTS

The Members of the working group are grateful to Smt C. Naga Rani I.A.S., Commissioner of Technical Education & Chairman of SBTET, for continuous guidance and valuable inputs during process of revising, modifying and updating the Curriculum C-20 to Curriculum C-23.

We are grateful to Sri. S. Suresh Kumar, I.A.S, Principal Secretary, Skills Development & Training for his valuable suggestions to bring the revamped curriculum C-23 in to a final form to meet latest Industry 4.0 concepts.

We are grateful to Sri. Saurab Gaur, I.A.S, former Principal Secretary, Skills Development & Training who actively participated in the Industry-Academia workshop conducted on 15<sup>th</sup> February, 2023 and offered valuable suggestions and insights into the learning needs and preferences so that the curriculum is engaging, inclusive, and effective.

It is pertinent to acknowledge the support of the following in the making of Curriculum C-23. A series of workshops in different phases were conducted by SBTET, AP, Guntur involving faculty from Polytechnics, Premier Engineering Colleges & representatives from various Industries and Dr. C. R. Nagendra Rao, Professor & Head, NITTTR-ECV to analyse the Previous C-20 Curriculum and in designing of C-23 Curriculum, is highly appreciated and gratefully acknowledged.

We also extend our sincere thanks to Sri K.V. Ramana Babu, Secretary, SBTE&T, Andhra Pradesh, Sri K.Vijaya Bhaskar, Former Secretary, SBTET, Andhra Pradesh, Sri. V.Padma Rao, Joint Director of Technical Education, officials of Directorate of Technical Education and the State Board of Technical Education, Andhra Pradesh and all teaching fraternity from the Polytechnics who are directly or indirectly involved in preparation of the curricula.

## **4. RULES AND REGULATIONS OF C-23 CURRICULUM**

### **4.1 Duration and pattern of the courses**

All the Diploma programs run at various institutions are of AICTE approved 3 years or 3½ years duration of academic instruction. All the Diploma courses are run on year wise pattern in the first year, and the remaining two or two & half years are run in the semester pattern. In respect of few courses like Diploma in Bio-Medical course, the training will be in the seventh semester. **Run-through system is adopted for all the Diploma Courses, subject to eligibility conditions.**

### **4.2 Procedure for Admission into the Diploma Courses:**

Selection of candidates is governed by the Rules and Regulations laid down in this regard from time to time.

- a) Candidates who wish to seek admission in any of the Diploma courses will have to appear for the Common Entrance Test for admissions into Polytechnics (POLYCET) conducted by the State Board of Technical Education and Training, Andhra Pradesh, Vijayawada. Only the candidates satisfying the following requirements will be eligible to appear for the Common Entrance Test for admissions into Polytechnics (POLYCET).
  - a. The candidates seeking admission should have appeared for S.S.C examination, conducted by the Board of Secondary Education, Andhra Pradesh or equivalent examination thereto, at the time of applying for the Common Entrance Test for admissions into Polytechnics (POLYCET). In case of candidates whose results of their Qualifying Examinations is pending, their selection shall be subject to production of proof of their passing the qualifying examination in one attempt or compartmentally at the time of admission.
  - b. Admissions are made based on the merit obtained in the Common Entrance Test (POLYCET) and the reservation rules stipulated by the Government of Andhra Pradesh from time to time.
  - c. For admission into the following Diploma Courses for which entry qualification is 10+2, candidates need not appear for POLYCET. A separate notification will be issued for admission into these courses.
    - i). D.HMCT ii).D. Pharmacy

### **4.3 Medium of Instruction**

The medium of instruction and examination shall be English.

### **4.4 Permanent Identification Number (PIN)**

A cumulative / academic record is to be maintained of the Marks secured in sessional work and end examination of each year for determining the eligibility for promotion etc., A Permanent Identification Number (PIN) will be allotted to each admitted candidate to maintain academic records.

### **4.5 Number of Working Days Per Semester / Year:**

- a) The Academic year for all the Courses shall be in accordance with the Academic Calendar.
- b) The Working days in a week shall be from Monday to Saturday

- c) There shall be 7 periods of 50 minutes duration each on all working days.
- d) The minimum number of working days for each semester / year shall be 90 / 180 days excluding examination days. If this prescribed minimum is not achieved due to any reason, special arrangements shall be made to conduct classes to complete the syllabus.

#### 4.6 Eligibility (Attendance to Appear for the End Examination)

- a) A candidate shall be permitted to appear for the end examination in all subjects, if he or she has attended a minimum of 75% of working days during the year/Semester.
- b) Condonation of shortage of attendance in aggregate up to 10% (65% and above and below 75%) in each semester or 1st year may be granted on medical grounds.
- c) A stipulated fee shall be payable towards condonation for shortage of attendance.
- d) Candidates having less than 65% attendance shall be detained.
- e) Students whose shortage of attendance is not condoned in any semester / 1<sup>st</sup> year and not paid the condonation fee in time are not eligible to take their end examination of that class and their admissions shall stand cancelled. They may seek re-admission for that semester / 1<sup>st</sup> year when offered in the next subsequent academic semester/year.

#### For INDUSTRIAL TRAINING:

- i) During Industrial Training the candidate shall put in a minimum of 90% attendance.
- ii) If the student fails to secure 90% attendance during industrial training, the student shall reappear for 6 months industrial training at his own expenses.

#### 4.7 Readmission

Readmission shall be granted to eligible candidates by the respective Principal/ Regional Joint Director.

- a) (i) Within 15 days after commencement of class work in any semester (Except Industrial Training).
- (ii) For Industrial Training: before commencement of the Industrial training.
- b) Within 30 days after commencement of class work in any year (including D. Pharmacy course or first year course in Engineering and Non-Engineering Diploma streams). Otherwise, such cases shall not be considered for readmission for that semester / year and are advised to seek readmission in the next subsequent eligible academic year.
- c) The percentage of attendance of the readmitted candidates shall be calculated from the first day of beginning of the regular class work for that year / Semester, as officially announced by CTE/SBTET but not from the day on which he/she has actually reported to the class work.

#### 4.8 Scheme of Evaluation

##### a) First Year

**Theory Courses:** Each Course carries Maximum marks of 80 with an end examination of 3 hours duration, along with internal assessment for Maximum of

20 marks. (Sessional marks). However, there are no minimum marks prescribed for sessionals.

**Laboratory Courses:** There shall be 40/20 Marks for internal assessment i.e. sessional marks for each practical Course with an end examination of 3 hours duration carrying 60/30 marks. However, there are no minimum marks prescribed for sessional.

**b) III, IV, V, VI and VII Semesters:**

**Theory Courses:** End semester evaluation shall be of 3 hours duration and for a maximum of 80 marks.

**Laboratory Courses:** Each Course carry 60/30 marks of 3 hours duration 40/20 sessional marks.

**4.9 Internal Assessment Scheme**

**a) Theory Courses:** Internal assessment shall be conducted for awarding Sessional marks on the dates specified. **Three-unit tests shall be conducted for I year students and two Unit Tests for semesters. The details are presented below.**

S. No.	Type of Assessment	Weightage Assigned
(i)	Testing of knowledge through mid-examination for year/ sem as (Mid-1+Mid-2+Mid3) or (Mid-1 + Mid-2)	40
(ii)	Assignments	5
(iii)	<i>Dynamic Learning activities : Project Work/ Seminar/Tech-fest/Group Discussion, Quizzes etc./Extra-curricular activities/NSS/NCC/IPSGM/Cleaning &amp; Greening of Campus etc.</i>	5
	TOTAL	50

Internal Assessment shall be of 90 minutes duration and for a maximum of 40 marks for each test.

At least one assignment should be completed for each unit which carries 10 marks. The total assignment marks should be reduced to 5.

The dynamic learning activity is to be conducted which carries 10 marks. The total marks should be reduced to 5.

The total 50 marks assigned to internal assignment is to be scaled down to 20 marks.

**b) Practical Courses:**

**(i) Drawing Courses:**

The award of Sessional marks for internal Assessment shall be as given in the following table:

Distribution of Marks for the Internal Assessment Marks			
First Year (Total:40 Marks)		Semesters (Total:40 Marks)	
Max:20 Marks	Max:20 Marks	Max:20 Marks	Max:20 Marks
From the Average of THREE Unit Tests.	From the Average of Assessment of Regular Class work Exercises.	From the Average of TWO Unit Tests.	From the Average of Assessment of Regular Class work Exercises.

- For first year engineering drawing each unit test will be conducted for a duration of 2 hours with maximum marks of 40.
- (Part - A: 4 questions x 5 marks = 20 Marks; Part -B: 2 questions x 10 marks = 20 marks).
- For the semester drawing examinations, Two Unit tests shall be conducted as per the Board End Examination Question Paper Pattern.
- All Drawing exercises are to be filed in serial order and secured for further scrutiny by a competent authority

**(ii) Laboratory Courses:**

- (a) Student's performance in Laboratories / Workshop shall be assessed during the year/ semester of study for 40 marks in each practical Course.
- (b) Evaluation for Laboratory Courses, other than Drawing courses:
  - i. Instruction (teaching) in laboratory courses (except for the course on Drawing) here after shall be task/competency based as delineated in the Laboratory sheets, prepared by SBTET, AP & NITTTR- ECV and posted in SBTET website.
  - ii. Internal assessment for Laboratory shall be done on the basis of task/s performed by the student as delineated in the laboratory sheets, prepared by SBTET, AP & NITTTR- ECV and posted in AP, SBTET website.
  - iii. Question paper for End semester Evaluation shall also be task/s based and shall be prepared and distributed by SBTET as done in case of theory courses be prepared as per SBTET rules in vogue.
- c) Internal assessment in Labs / workshops / Survey field work etc., during the course of study shall be done and sessional marks shall be awarded by the concerned Teacher.
- d) For practical examinations, except in drawing, there shall be two examiners. External examiner shall be appointed by the Principal in consultation with respective Head of Section preferably choosing a qualified person from in the order of preference.
  - i) Nearby Industry
  - ii) Govt / Semi Govt organization like R & B, PWD, PR, Railways, BSNL, APSRTC, APSEB etc.
  - iii) Govt / University Engg College.
  - iv) HoD/Senior Lecture ( Selection Grade-II) from the Govt. Polytechnic

Internal examiner shall be the person concerned with internal assessment as in (c) above. The end examination shall be held along with all theory papers in respect of drawing.

- e) Question Paper for Practicals: Question paper should cover ( the experiments / exercise prescribed to test various) skills like handling, manipulating, testing, trouble shooting, repair, assembling and dismantling etc., from more than one experiment / exercise
- f) Records pertaining to internal assessment marks of both theory and practical Courses are to be maintained for official inspection.
- g) In case of Diploma programs having Industrial Training, Internal Assessment and Summative Evaluation, shall be done as illustrated in the following table:

Assessment no	Upon completion of	By	Based on	Max Marks
1	12 weeks	1.The faculty concerned (Guide) and 2. Training in charge (Mentor) of the industry	Learning outcomes as given in the scheme of assessment ,for Industrial Training	120
2	22 weeks			120
3. Final summative Evaluation	24 week	1.The faculty member concerned, 2.HoD concerned and 3.An external examiner	1.Demonstration of any one of the skills listed in learning outcomes	30
			2.Training Report	20
			3.Viva Voce	10
<b>TOTAL</b>				<b>300</b>

- h) Each staff member including Head of Section shall be assigned a batch of students 10 to 15 for making assessment during industrial training.

#### 4.10 Minimum Pass Marks

##### a) Theory Examination:

For passing a theory Course, a candidate has to secure a minimum of 35% in end examination and a combined minimum of 35% of both Sessional and end examination marks put together.

##### b) Practical Examination:

For passing a practical Course, a candidate has to secure a minimum of 50% in end examination and a combined minimum of 50% of both sessional and practical end

examination marks put together. In case of D.C.C.P., the pass mark for typewriting and short hand is 45% in the end examination. There are no sessional marks for typewriting and Shorthand Courses of D.C.C.P course.

**C) Industrial Training:**

- I. Monitoring: Similar to project work each teacher may be assigned a batch of 10-15 students irrespective of the placement of the students to facilitate effective monitoring of students learning during industrial training.
- II. Assessment: The Industrial training shall carry 300 marks and pass marks is 50% in assessments at industry (first and second assessment) and final summative assessment at institution level put together i.e. 150 marks out of 300 marks. And also student has to secure 50% marks in final summative assessment at institution level.
- III. **In-Plant Industrial Training for 3-Year Diploma (C-23) Courses is scheduled as per the Academic Calendar of the SBTET every year.**

**4.11. Provision for Improvement**

Improvement is allowed only after he / she has completed all the Courses from First Year to Final semester of the Diploma.

- a) Improvement is allowed in any 4 (Four) Courses of the Diploma.
- b) The student can avail of this improvement chance **ONLY ONCE**, that too within the succeeding two examinations after the completion of Diploma. However, the duration including Improvement examination shall not exceed **FIVE** years from the year of first admission.
- c) No improvement is allowed in Practical / Lab Courses or Project work or Industrial Training assessment. However, improvement in drawing Course(s) is allowed.
- d) If improvement is not achieved, the marks obtained in previous Examinations hold good.
- e) Improvement is not allowed in respect of the candidates who are punished under Mal-practice in any Examination.
- f) Examination fee for improvement shall be paid as per the notification issued by State Board of Technical Education and Training from time to time.
- g) All the candidates who wish to appear for improvement of performance shall deposit the original Marks Memos of all the years / Semesters and also original Diploma Certificate to the Board. If there is improvement in performance of the current examination, the revised Memorandum of marks and Original Diploma Certificate will be issued, else the submitted originals will be returned.

**4.12. Rules of Promotion From 1<sup>ST</sup> YEAR TO 3<sup>rd</sup>, 4<sup>th</sup>, 5<sup>th</sup>, 6<sup>th</sup> and 7<sup>th</sup> Semesters:**

**A) For Diploma Courses of 3 Years duration**

- i). A candidate shall be permitted to appear for first year examination provided he / she puts in 75% attendance (which can be condoned on Medical grounds up to 10%) and pay the examination fee.
- ii) A candidate shall be promoted to 3rd semester if he/she puts the required percentage of attendance in the first year and pays the examination fee. A candidate who could not pay the first year examination fee has to pay the

promotion fee as prescribed by State Board of Technical Education and Training, AP from time to time before commencement of 3rd semester.

A candidate is eligible to appear for the 3rd semester examination if he/she puts the required percentage of attendance in the 3rd semester and pays the examination fee.

iii) A candidate shall be promoted to 4th semester provided he/she puts the required percentage of attendance in the 3rd semester and pay the examination fee. A candidate, who could not pay the 3rd semester exam fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training AP from time to time before commencement of 4th semester. A candidate is eligible to appear for the 4th semester examination if he/she puts the required percentage of attendance in the 4th semester and pays the examination fee.

iv) A candidate shall be promoted to 5<sup>th</sup> semester provided he / she puts the required percentage of attendance in the 4<sup>th</sup> semester and pays the examination fee. A candidate, who could not pay the 4<sup>th</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 5<sup>th</sup> semester.

A candidate is eligible to appear for the 5<sup>th</sup> semester examination if he/she puts the required percentage of attendance in the 5<sup>th</sup> semester and pays the examination fee.

v) A candidate shall be sent to Industrial training / VI semester provided he/she puts in the required percentage of attendance in the 5<sup>th</sup> semester and pay the examination fee/ promotion fee as prescribed by SBTET.  
A candidate is eligible to appear for Industrial Training assessment (Seminar/Viva-voce) puts the required percentage of attendance, i.e., 90% in 6th semester Industrial Training.

**For IVC & ITI Lateral Entry students:**

i.) A candidate shall be permitted to appear for Third Semester examination provided he / she puts in 75% attendance (which can be condoned on Medical grounds up to 10%) and pay the examination fee for Third semester.

ii) A candidate shall be promoted to 4th semester provided he/she puts the required percentage of attendance in the 3rd semester and pay the examination fee. A candidate, who could not pay the 3rd semester exam fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training AP from time to time before commencement of 4th semester.

A candidate is eligible to appear for the 4th semester examination if he/she puts the required percentage of attendance in the 4th semester and pays the examination fee.

ii) A candidate shall be promoted to 5<sup>th</sup> semester provided he / she puts the required percentage of attendance in the 4<sup>th</sup> semester and pays the examination fee. A candidate, who could not pay the 4<sup>th</sup> semester examination fee, has to pay the

promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 5<sup>th</sup> semester.

A candidate is eligible to appear for the 5<sup>th</sup> semester examination if he/she puts the required percentage of attendance in the 5<sup>th</sup> semester and pays the examination fee.

- iii) A candidate shall be sent to Industrial training / VI semester provided he/she puts in the required percentage of attendance in the 5<sup>th</sup> semester and pay the examination fee/ promotion fee as prescribed by SBTET.

A candidate is eligible to appear for Industrial Training assessment (Seminar/Viva-voce) puts the required percentage of attendance, i.e., 90% in 6<sup>th</sup> semester Industrial Training and pays the examination fee.

**B) For Diploma Courses of 3 ½ Years duration (MET/ CH/ CHPP/ CHPC/ CHOT/ TT):**

- i. A candidate shall be permitted to appear for 1<sup>st</sup> year examination provided he / she puts in 75% attendance (which can be condoned on Medical grounds upto 10%) i.e. attendance after condonation on Medical grounds should not be less than 65% and pay the examination fee.
- ii. A candidate shall be promoted to 3<sup>rd</sup> semester if he/she puts the required percentage of attendance in the 1<sup>st</sup> year and pays the examination fee. A candidate who could not pay the 1<sup>st</sup> year examination fee has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 3<sup>rd</sup> semester.
- iii. A candidate shall be promoted to 4<sup>th</sup> semester provided he/she puts the required percentage of attendance in the 3<sup>rd</sup> semester and pay the examination fee. A candidate, who could not pay the 3<sup>rd</sup> semester exam fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 4<sup>th</sup> semester.
- A candidate is eligible to appear for the 4<sup>th</sup> semester exam if he/she puts the required percentage of attendance in the 4<sup>th</sup> semester

**For IVC & ITI Lateral Entry students:**

- a) Puts the required percentage of attendance in the 4<sup>th</sup> semester
- iv. A candidate shall be promoted to 5<sup>th</sup> semester industrial training provided he / she puts the required percentage of attendance in the 4<sup>th</sup> semester and pays the examination fee. A candidate, who could not pay the 4<sup>th</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 5<sup>th</sup> semester.
- v. Promotion from 5<sup>th</sup> to 6<sup>th</sup> semester is automatic (i.e., from 1<sup>st</sup> spell of Industrial Training to 2<sup>nd</sup> spell) provided he/she puts the required percentage of attendance, which in this case i.e., 90 % of attendance and attends for the VIVA-VOCE examination at the end of training.
- vi. A candidate shall be promoted to 7<sup>th</sup> semester provided he / she puts the required percentage of attendance in the 6<sup>th</sup> semester and pays the examination fee. A candidate, who could not pay the 6<sup>th</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 7<sup>th</sup> semester.
- vii. A candidate shall be promoted to 7<sup>th</sup> semester of the course provided he/she has successfully completed both the spells of Industrial Training.

A candidate is eligible to appear for 7th semester examination if he/she

- a) Puts in the required percentage of attendance in the 7th semester

**For IVC & ITI Lateral Entry students:**

- a) Puts in the required percentage of attendance in the 7th semester .

**C) For Diploma Courses of 3 ½ Years duration (BM):**

The same rules which are applicable for conventional courses also apply for this course. The industrial training in respect of this course is restricted to one semester (6 months) after the 6<sup>th</sup> semester (3 years) of the course.

- i. A candidate shall be permitted to appear for first year examination provided he / she puts in 75% attendance (which can be condoned on Medical grounds upto 10%) i.e. attendance after condonation on Medical grounds should not be less than 65% and pay the examination fee.
- ii. A candidate shall be promoted to 3<sup>rd</sup> semester if he/she puts the required percentage of attendance in the first year and pays the examination fee. A candidate who could not pay the first year examination fee has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 3<sup>rd</sup> semester.
- iii. A candidate shall be promoted to 4<sup>th</sup> semester provided he/she puts the required percentage of attendance in the 3<sup>rd</sup> semester and pay the examination fee. A candidate who could not pay the 3<sup>rd</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 4<sup>th</sup> semester.  
A candidate is eligible to appear for the 4<sup>th</sup> semester examination if he/she
  - a) Puts in the required percentage of attendance in the 4<sup>th</sup> semester

**For IVC & ITI Lateral Entry Students:**

A candidate is eligible to appear for the 4<sup>th</sup> semester examination if he/she puts the required percentage of attendance in the 4<sup>th</sup> semester

- iv. A candidate shall be promoted to 5<sup>th</sup> semester provided he / she puts the required percentage of attendance in the 4<sup>th</sup> semester and pays the examination fee. A candidate, who could not pay the 4<sup>th</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 5<sup>th</sup> semester.  
A candidate is eligible to appear for the 5<sup>th</sup> semester exam if he/she
  - a) Puts in the required percentage of attendance in the 5<sup>th</sup> semester.

**For IVC & ITI Lateral Entry students:**

- a) Puts in the required percentage of attendance in the 5<sup>th</sup> semester.
- v. A candidate shall be promoted to 6<sup>th</sup> semester provided he/she puts in the required percentage of attendance in the 5<sup>th</sup> semester and pays the examination fee.  
A candidate who could not pay the 5<sup>th</sup> semester examination fee, has to pay the promotion fee as prescribed by State Board of Technical Education and Training from time to time before commencement of 6<sup>th</sup> semester.  
A candidate is eligible to appear for 6<sup>th</sup> semester examination
  - a) Puts in the required percentage of attendance in 6<sup>th</sup> semester

**IVC & ITI Lateral Entry students:**

- a) Puts in the required percentage of attendance in 6<sup>th</sup> semester.
  
- vi. A candidate shall be promoted to 7<sup>th</sup> semester provided he/she puts in the required percentage of attendance in 6<sup>th</sup> semester and pay the examination fee. A candidate, who could not pay the 6<sup>th</sup> semester examination fee, has to pay the promotion fee prescribed by SBTET from time to time before commencement of the 7<sup>th</sup> semester (Industrial Training).

A candidate is eligible to appear for 7<sup>th</sup> semester Industrial Training assessment (Seminar/Viva-voce) if he/she

- a) Puts in the required percentage of attendance, i.e., 90% in 7<sup>th</sup> semester Industrial Training.

**For IVC & ITI Lateral Entry students:**

- a) Puts in the required percentage of attendance, i.e., 90% in 7<sup>th</sup> semester Industrial Training.

**4.13. Students Performance Evaluation**

Successful candidates shall be awarded the Diploma under the following divisions of pass.

- a) First Class with Distinction shall be awarded to the candidates who secure an overall aggregate of 75% marks and above.
- b) First Class shall be awarded to candidates who secure overall aggregate of 60% marks and above and below 75% marks.
- c) Second Class shall be awarded to candidates who secure a pass with an overall aggregate of below 60%.
  - i. The Weightage of marks for various year/Semesters which are taken for computing overall aggregate shall be 25% of I year marks + 100% of 3<sup>rd</sup> and subsequent Semesters.
  - ii. In respect IVC & ITI Lateral Entry candidates who are admitted directly into diploma course at the 3<sup>rd</sup> semester (i.e., second year) level the aggregate of (100%) marks secured at the 3<sup>rd</sup> and subsequent semesters of study shall be taken into consideration for determining the overall percentage of marks secured by the candidates for award of class/division.
- d) Second Class shall be awarded to all students, who fail to complete the Diploma in the regular 3 years/ 3 ½ years and four subsequent examinations from the year of first admission.

**4.14. EXAMINATION FEE SCHEDULE:**

The examination fee should be as per the notification issued by State Board of Technical Education and Training, AP from time to time.

**4.15. Structure of Examination Question Paper:**

**I. Formative assessment (Internal examination)**

- a) For theory Courses:

Three-unit tests for first year and two-unit tests for semesters shall be conducted with a duration of 90 minutes for each test for maximum marks of 40. It consists of part A and Part B.

**Part A** contains five questions and carries 16 marks. Among these five questions first question consists of four objective items like one word or phrase answer/filling-in the blanks/true or false etc with one mark for each question. The other four questions are short answer questions and carry three marks each.

**Part B** carries 24 marks and consists of three questions with internal choice i.e., Either/Or type, and each question carries 8 marks.

The sum of marks of 3 tests for I year and 2 tests for semesters including assignments and Dynamic learning activities (50 marks) shall be reduced to 20 marks in each Course for arriving at final sessional marks.

**b) For drawing Courses:**

**For I year:**

Three-unit tests with duration of 90 minutes and for maximum marks of 40 marks shall be conducted for first year. It consists of part A and Part B.

Part A consists four questions for maximum marks of 16 and each question carries four marks (4×4 marks=16 marks).

Part B carries maximum marks of 24 and consists of five questions while the student shall answer any three questions out of these five questions. Each question in this part carries a maximum mark of 8, (3×8 marks=24 marks).

The sum of marks obtained in 3-unit test marks shall be reduced to 20 marks for arriving at final sessional marks. Remaining 20 marks are awarded by the Course teacher based on the student's performance during regular class exercise.

**For semester:** Two-unit tests with duration of 90 minutes and for maximum marks of 40 marks shall be conducted. The sum of marks obtained in 2-unit test marks shall be reduced to 20 marks for arriving at final sessional marks. Remaining 20 marks are awarded by the Course teacher based on the student's performance during regular class exercise.

**c) For Laboratory /workshop:** 50% of total marks for the Course shall be awarded based on continuous assessment of the student in laboratory/workshop classes and the remaining 50% shall be based on the sum of the marks obtained by the students in two tests.

## **II. Summative assessment (End examination)**

The question paper for theory examination is patterned in such a manner that the Weightage of periods/marks allotted for each of the topics for a particular Course be considered. End Examination paper is of 3 hours duration.

- a) **Each theory paper consists of Section 'A' and 'B'**  
**Section 'A' with Max marks of 30**, contains 10 short answer questions. All questions are to be answered and each carry 3 marks, i.e.,  $10 \times 3 = 30$ .

**Section 'B' with Max marks of 50** contains 8 essay type questions. Only 5 questions are to be answered and each carry 10 marks, i.e., Max. Marks:  $5 \times 10 = 50$ .

**Thus, the total marks for theory examination shall be: 80.**

- b) **For Engineering Drawing Course (107) consist of section 'A' and section 'B'.**

**Section 'A' with max marks of 20**, contains four (4) questions. All questions in section 'A' are to be answered to the scale and each carries 5 marks, ie.  $4 \times 5=20$ .

**Section 'B' with max marks of 40**, contains six (6) questions. The student shall answer any four (4) questions out of the above six questions and each question carries 10 Marks, i.e.,  $4 \times 10 = 40$ .

- c) **Practical Examinations**

For Workshop practice and Laboratory Examinations, Each student has to pick up a question paper distributed by Lottery System.

Max. Marks for an experiment / exercise : 50

Max. Marks for VIVA-VOCE : 10

Total Max. Marks : 60

In case of practical examinations with 50 marks, the marks shall be distributed as

Max. Marks for an experiment / exercise : 25

Max. Marks for VIVA-VOCE : 05

Total Max. Marks : 30

In case of any change in the pattern of question paper, the same shall be informed sufficiently in advance to the candidates.

- d) **Note: Evaluation for Laboratory Courses, other than Drawing courses:**

- I. Instruction (teaching) in laboratory courses (except for the course on Drawing) hereafter shall be task/competency based as delineated in the Laboratory sheets, prepared by SBTET, AP and posted in its website.
- II. Internal assessment for Laboratory shall be done on basis of task/s performed by the student as delineated in the laboratory sheets, prepared by SBTET, AP and posted in its website.
- III. Question paper for End semester Evaluation shall be prepared as per SBTET rules in vogue.

#### 4.16. ISSUE OF MEMORONDUM OF MARKS

All candidates who appear for the end examination will be issued memorandum of marks without any payment of fee. However candidates who lose the original memorandum of marks have to pay the prescribed fee to the Secretary, State Board of Technical Education and Training, A.P. for each duplicate memo from time to time.

**4.17. MAXIMUM PERIOD FOR COMPLETION OF DIPLOMA PROGRAMMES:**

Maximum period for completion of the diploma courses is twice the duration of the course from the date of First admission (includes the period of detention and discontinuation of studies by student etc) failing which they will have to forfeit the claim for qualifying for the award of Diploma (They will not be permitted to appear for examinations after that date). This rule applies for all Diploma courses of 3 years and 3 ½ years of engineering and non-engineering courses.

**4.18. ELIGIBILITY FOR AWARD OF DIPLOMA**

A candidate is eligible for award of Diploma Certificate if he / she fulfil the following academic regulations.

- i. He / She pursued a course of study for not less than 3 / 3 ½ academic years & not more than 6 / 7 academic years.
- ii. He / she have completed all the Courses.  
Students who fail to fulfil all the academic requirements for the award of the Diploma within 6 / 7 academic years from the year of admission shall forfeit their seat in the course & their seat shall stand cancelled.

**For IVC & ITI Lateral Entry students:**

- i. He / She pursued a course of study for not less than 2 / 2 ½ academic years & not more than 4 / 5 academic years.
- ii. He / she has completed all the Courses.  
Students who fail to fulfil all the academic requirements for the award of the Diploma within 4 / 5 academic years from the year of admission shall forfeit their seat in the course & their seat shall stand cancelled.

**4.19. ISSUE OF PHOTO COPY OF VALUED ANSWER SCRIPT, RECOUNTING & REVERIFICATION:**

**A) FOR ISSUE OF PHOTO COPIES OF VALUED ANSWER SCRIPTS**

- I. A candidate desirous of applying for Photo copy of valued answer script/s should apply within prescribed date from the date of the declaration of the result.
- II. Photo copies of valued answer scripts will be issued to all theory Courses and Drawing Course (s).
- III. The Photo copy of valued answer script will be dispatched to the concerned candidate's address as mentioned in the application form by post.
- IV. No application can be entertained from third parties.

**B) FOR RE-COUNTING (RC) and RE-VERIFICATION(RV) OF THE VALUED ANSWER SCRIPT**

- i. A candidate desirous of applying for Re-verification of valued answer script should apply within prescribed date from the date of the declaration of the result.
- ii. Re-verification of valued answer script shall be done for all theory Courses' and Drawing Course(s).
- iii. The Re-verification committee constituted by the Secretary, SBTETAP with Course experts shall re-verify the answer scripts.

I. **RE-COUNTING**

The Officer of SBTET will verify the marks posted and recount them in the already valued answer script. The variations if any will be recorded separately, without making any changes on the already valued answer script. The marks awarded in the original answer script are maintained (hidden).

II. **RE-VERIFICATION**

- (i) The Committee has to verify the intactness and genuineness of the answer script(s) placed for Re-verification.
- (ii) Initially single member shall carry out the re-verification.
- (iii) On re-verification by single member, if the variation is less than 12% of maximum marks, and if there is no change in the STATUS in the result of the candidate, such cases will not be referred to the next level i.e., for 2-Tier evaluation.
- (iv) On re-verification by a single member, if the variation is more than 12% of maximum marks, it will be referred to 2-Tier evaluation.
- (v) If the 2-Tier evaluation confirms variation in marks as more than 12% of maximum marks, the variation is considered as follows:
  - a) If the candidate has already passed and obtains more than 12% of the maximum marks on Re-verification, then the variation is considered.
  - b) If the candidate is failed and obtains more than 12% of the maximum marks on Re-verification and secured pass marks on re-verification, then the status of the candidate changes to PASS.
  - c) If a candidate is failed and obtains more than 12% of the maximum marks on Re-verification and if the marks secured on re-verification are still less than the minimum pass marks, the status of the candidate remain FAIL only.
- (vii) After Re-verification of valued answer script the same or change if any therein on Re-verification, will be communicated to the candidate.
- (viii) On Re-verification of Valued Answer Script if the candidate's marks are revised, the fee paid by the candidate will be refunded or else the candidate has to forfeit the fee amount.

**Note:** No request for Photo copies/ Recounting /Re-verification of valued answer script would be entertained from a candidate who is reported to have resorted to Malpractice in that examination.

**4.20. Mal Practice Cases:**

If any candidate resorts to Mal Practice during examinations, he / she shall be booked and the Punishment shall be awarded as per SBTETAP rules and regulations in vogue.

**4.21. Discrepancies/ Pleas:**

Any Discrepancy /Pleas regarding results etc., shall be represented to the SBTETAP within one month from the date of issue of results. Thereafter, no such cases shall be entertained in any manner.

**4.22. Issue of Duplicate Diploma**

If a candidate loses his/her original Diploma Certificate and desires a duplicate to be issued he/she should produce written evidence to this effect. He / she may obtain a

duplicate from the Secretary, State Board of Technical Education and Training, A.P., on payment of prescribed fee and on production of an affidavit signed before a First Class Magistrate (Judicial) and non-traceable certificate from the Department of Police. In case of damage of original Diploma Certificate, he / she may obtain a duplicate certificate by surrendering the original damaged certificate on payment of prescribed fee to the State Board of Technical Education and Training, A.P.

In case the candidate cannot collect the original Diploma within 1 year from the date of issue of the certificate, the candidate has to pay the penalty prescribed by the SBTET AP from time to time.

#### **4.23. Issue of Migration Certificate and Transcripts:**

The Board on payment of prescribed fee will issue these certificates for the candidates who intend to prosecute Higher Studies in India or Abroad.

#### **4.24. General**

- i. The Board may change or amend the academic rules and regulations or syllabi at any time and the changes or amendments made shall be applicable to all the students, for whom it is intended, with effect from the dates notified by the competent authority.
- ii. All legal matters pertaining to the State Board of Technical Education and Training, AP are within the jurisdiction of Mangalagiri.
- iii. In case of any ambiguity in the interpretation of the above rules, the decision of the Secretary, SBTET, A.P is final.

## C-23 Curriculum for DEEE

### With Industrial training (In-house) in Semester VI

#### **VISION**

To develop Electrical & Electronics Engineering professionals competent to face the global challenges in a Edifying environment conducive to learn technical knowledge, skills blended with ethics and values, to Coordinate and serve to the society for betterment and comfortable living.

#### **MISSION**

M1	To provide a competitive learning environment, through a need based curriculum designed in collaboration with industry, conducive for high quality education emphasising on transfer of knowledge and skill development essential for the profession and the society as well.
M2	To nurture higher order leadership qualities and ethics and values in students to enable them to be leaders in their chosen professions while maintaining the highest level of ethics.
M3	To encourage the spirit of inquisition to promote innovation and entrepreneurship strengthened with life skills to sustain the stress.
M4	To foster effective interactions and networking with all the stake holders so as to work towards the growth and sustainability of the society and environment.

#### **Programme Educational Objectives (PEOs)**

On completion of the Diploma Electrical & Electronics Engineering programme, the students should have acquired the following characteristics

PEO1	An ability to apply knowledge of mathematics, Science , engineering and management principles in solving problems in the field of Electrical and Electronics Engineering.
PEO2	To be life-long learners with sprit of enquiry and zeal to acquire new knowledge and skills so as to remain contemporary and posses required professional skills.
PEO3	To enhance entrepreneurial, communication and other soft skills, which will enable them to work globally as leaders, team members and contribute to nation building for the betterment of the society.
PEO4	To make them strongly committed to the highest levels of professional ethics and focus on ensuring quality, adherence to public policy and law, safety, reliability and environmental sustainability in all their professional activities

## PROGRAMME OUTCOMES(POs)

1. **Basic and discipline specific knowledge:** Apply knowledge of basic mathematics, science and engineering fundamentals and engineering specialization to solve the engineering problems.
2. **Problem analysis:** Identify and analyse well-defined engineering problems using standard methods
3. **Design/Development of solutions:** Design solutions for well-defined technical problems and assist with the design of systems components or processes to meet specified needs
4. **Engineering tools, Experimentation and Testing:** Apply modern engineering tools and appropriate technique to conduct standard tests and measurements.
5. **Engineering practices for society, sustainability and environment:** Apply appropriate technology in context of society, sustainability, environment and ethical practices.
6. **Project Management:** Use engineering management principles individually, as a team member or a leader to manage projects and effectively communicate about well defined engineering activities.
7. **Life-Long learning:** Ability to analyse individual needs and engaging updating in the context of technological changes.

## PROGRAMME SPECIFIC OUTCOMES (PSOs)

1. An ability to understand the basic concepts of Electrical & Electronics Engineering and to apply them to various areas like Wiring Installations, Lighting Schemes , Static & Rotating machinery, drawing layouts , Power System ( Generation, Transmission, Distribution & utilisation), Digital electronics, power control devices, Computer programming ,managerial skills and the use SMART technologies .
2. An ability to Repair, develop and troubleshooting of Various Electrical & Electronics equipment's by using suitable tools and techniques, to design Customized applications in Electrical & Electronics Engineering at economic and efficient considerations, to develop software & hardware solutions.
3. Wisdom of social and environmental awareness along with ethical responsibility to have a successful career and to sustain passion and zeal in the field of Electrical & Electronics Engineering for real-world applications in the field of Electronics using optimal resources as an Entrepreneur.

# **FIRST YEAR**

**DIPLOMA IN ELECTRICAL & ELECTRONICS ENGINEERING**  
**SCHEME OF INSTRUCTIONS AND EXAMINATIONS**

(FIRST YEAR) Subject Code	Name of the Subject	Instruction periods/week		Total Periods /year	Scheme of Examination			
		Theory	Practical/ Tutorial		Duration (hours)	Sessional Marks	End Exam Marks	Total Marks
<b>THEORY</b>								
EE-101	English	3	-	90	3	20	80	100
EE-102	Engineering Mathematics - I	5	-	150	3	20	80	100
EE-103	Engineering Physics	3	-	90	3	20	80	100
EE-104	Engineering chemistry & Environmental studies	3	-	90	3	20	80	100
EE-105	Electrical Engineering Material Science	4	-	120	3	20	80	100
EE-106	Basic Electrical Technology	6	-	180	3	20	80	100
<b>PRACTICAL</b>								
EE-107	Engineering Drawing	-	3	90	3	40	60	100
EE-108	Electrical Wiring Laboratory	-	6	180	3	40	60	100
EE-109	Physics Lab	-	1.5	45	1½	20	30	50
EE-110	Chemistry Lab	-	1.5	45	1½	20	30	50
EE-111	Computer Fundamentals Laboratory	-	3	90	3	40	60	100
	<b>TOTAL</b>	<b>24</b>	<b>15</b>	<b>1170</b>	<b>30</b>	<b>280</b>	<b>720</b>	<b>1000</b>
<b>NOTE:</b> 03 periods per week are allotted to Student Centric Activity (Library, Sports& Games, Clean & Green, Preparation for placements etc)								

**ENGLISH**

Course Code	Course Title	No. of Periods per Week	Total No. of Periods	Marks for FA	Marks for SA
EE-101	English	3	90	20	80

Time Schedule : EE- 101 : ENGLISH						
Chapter No.	Title	No. of Periods	Weightage	No. of short questions	No. of Essay questions	CO'S Mapped
1	English for Employability	8	16	2	1	CO1, CO2, CO3, CO4, CO5
2	Living in Harmony	8				CO1, CO2, CO3, CO4, CO5
3	Connect with Care	8	26	2	2	CO1, CO2, CO3, CO4, CO5
4	Humour for Happiness	8				CO1, CO2, CO3, CO4, CO5
5	Never Ever Give Up!	8	10	1	1	CO1, CO2, CO3, CO4, CO5
6	Preserve or Perish	9	23			2
7	The Rainbow of Diversity	8		CO1, CO2, CO3, CO4, CO5		
8	New Challenges -Newer Ideas	8	19	2	1	CO1, CO2, CO3, CO4, CO5
9	The End Point First	8				CO1, CO2, CO3, CO4, CO5
10	The Equal Halves	8	16	1	1	CO1, CO2, CO3, CO4, CO5
11	Dealing with Disaster	9				CO1, CO2, CO3, CO4, CO5
	<b>Total</b>	<b>90</b>	<b>110</b>	<b>30</b>	<b>80</b>	

S. No.	Title of the Unit	No of Periods	COs Mapped
1	English for Employability	8	CO1, CO2, CO3, CO4,CO5
2	Living in Harmony	8	CO1, CO2, CO3, CO4,CO5
3	Connect with Care	8	CO1, CO2, CO3, CO4, CO5
4	Humour for Happiness	8	CO1, CO2, CO3, CO4, CO5
5	Never Ever Give Up!	8	CO1, CO2, CO3, CO4, CO5
6	Preserve or Perish	9	CO1, CO2, CO3, CO4, CO5
7	The Rainbow of Diversity	8	CO1, CO2, CO3, CO4, CO5
8	New Challenges- Newer Ideas	8	CO1, CO2, CO3, CO4, CO5
9	The End Point First	8	CO1, CO2, CO3, CO4, CO5
10	The Equal Halves	8	CO1, CO2, CO3, CO4, CO5
11	Dealing with Disaster	9	CO1, CO2, CO3, CO4, CO5
<b>Total Periods</b>		<b>90</b>	

<b>COURSE OBJECTIVES</b>	- To improve grammatical knowledge and enrich vocabulary.
	- To develop effective reading, writing and speaking skills.
	- To comprehend themes related to Personality, Society, Environment to exhibit Universal Human Values.

<b>COURSE OUTCOMES</b>	CO1	Learn and apply various grammatical concepts to communicate in academic, professional and everyday situations.
	CO2	Use appropriate vocabulary in various contexts.
	CO3	Read and comprehend different forms of academic, professional and general reading material.
	CO4	Communicate effectively in speaking and writing in academic, professional and everyday situations.
	CO5	Display human values by applying the knowledge of themes related to Self, Society, Environment, Science and Technology for holistic development and harmonious living through communication.

### CO-PO Matrix

Course Code EE-101	Course Title: English Number of Course Outcomes: 5				No. of Periods: 90
POs	Mapped CO No.	CO Periods Addressing PO in Column 1		Level of Mapping (1,2,3)	Remarks
		Number	Percentage		
PO1		Not directly Applicable for English course, however, the language activities make use of the content from Science and Technology relevant to the programme to enhance English communication skills.			
PO2					
PO3					
PO4					
PO5	CO5	16	18%	Level 1	Up to 20%: Level 1 21%-50%: Level 2 >50%: Level 3
PO6	CO1, CO2, CO3, CO4,	52	58%	Level 3	
PO7	CO1, CO2, CO3, CO4,CO5	22	24%	Level 2	

Level 3 – Strongly Mapped, Level 2- Moderately Mapped; Level 1- Slightly Mapped

### LEARNING OUTCOMES

#### 1. English for Employability

- 1.1. Perceive the need for improving communication in English for employability
- 1.2. Use adjectives and articles effectively while speaking and in writing
- 1.3. Write simple sentences

#### 2. Living in Harmony

- 2.1. Develop positive self-esteem for harmonious relationships
- 2.2. Use affixation to form new words
- 2.3. Use prepositions and use a few phrasal verbs contextually

#### 3. Connect with Care

- 3.1. Use social media with discretion
- 3.2. Speak about abilities and possibilities
- 3.3. Make requests and express obligations
- 3.4. Use modal verbs and main verbs in appropriate form

3.5. Write short dialogues about everyday situations

#### **4. Humour for Happiness**

- 4.1. Realize the importance of humour for a healthy living
- 4.2. Improve vocabulary related to the theme
- 4.3. Inculcate reading and speaking skills
- 4.4. Frame sentences with proper Subject – Verb agreement
- 4.5. Understand the features of a good paragraph and learn how to gather ideas as a preliminary step for writing a good paragraph.

#### **5. Never Ever Give Up!**

- 5.1. Learn to deal with failures in life
- 5.2. Use the present tense form for various every day communicative functions such as speaking and writing about routines, professions, scientific descriptions and sports commentary
- 5.3. Write paragraphs with coherence and other necessary skills

#### **6. Preserve or Perish**

- 6.1. Understand the ecological challenges that we face today and act to save the environment.
- 6.2. Narrate / Report past events and talk about future actions
- 6.3. Develop vocabulary related to environment
- 6.4. Write e-mails

#### **7. The Rainbow of Diversity**

- 7.1. Appraise and value other cultures for a happy living in multi-cultural workspace
- 7.2. Understand the usage of different types of sentences
- 7.3. Ask for or give directions, information, instructions
- 7.4. Use language to express emotions in various situations
- 7.5. Write letters in various real life situations

#### **8. New Challenges – Newer Ideas**

- 8.1. Understand the functional difference between Active Voice and Passive Voice
- 8.2. Use Passive Voice to speak and write in various contexts
- 8.3. Understand the major parts and salient features of an essay
- 8.4. Learn about latest innovations and get motivated

#### **9. The End Point First!**

- 9.1. Understand the importance of setting goals in life
- 9.2. Report about what others have said both in speaking and writing
- 9.3. Write an essay following the structure in a cohesive and comprehensive manner
- 9.4. Apply the words related to Goal Setting in conversations and in life

#### **10. The Equal Halves**

- 10.1. Value the other genders and develop a gender-balanced view towards life
- 10.2. Identify the use of different conjunctions in synthesising sentences
- 10.3. Write various types of sentences to compare and contrast the ideas
- 10.4. Apply the knowledge of sentence synthesis in revising and rewriting short essays
- 10.5. Develop discourses in speech and writing

## 11. Dealing with Disasters

- 11.1. be aware of different kinds of disasters and the concept of disaster management
- 11.2. Generate vocabulary relevant to disaster management and use it in sentences
- 11.3. Analyze an error in a sentence and correct it
- 11.4. Learn and write different kinds of reports

Textbook: **'INTERACT'** (A Text book of English for I Year Engineering Diploma Courses) - by SBTET, AP

### Reference Books:

1. Martin Hewings: *Advanced Grammar in Use*, Cambridge University Press
2. Murphy, Raymond : *English Grammar in Use*, Cambridge University Press
3. Sidney Greenbaum : *Oxford English Grammar*, Oxford University Press
4. Wren and Martin (Revised by N.D.V. Prasad Rao) : *English Grammar and Composition*, Blackie ELT Books, S. Chand and Co.
5. Sarah Freeman: *Strengthen Your Writing*, Macmillan

	<b>End Exam ( 80 Marks)</b>	<b>1,2,3 Unit Tests ( 20 Marks each)</b>
Part A	10 Question @ 3 Marks	5 Questions @ (1Q X4M) + (4Q X3M =12)
	Total = 30 Marks	Total = 16 Marks
Part B	5 Questions ( + 3 Choice) @10 Marks	3 Questions ( with internal choice) @ 8 Marks
	Total = 50 Marks	Total = 24 marks
<b>Grand Total</b>	<b>80 Marks</b>	<b>40 Marks</b>

**ENGINEERING MATHEMATICS-I**  
(Common to all Branches)

Course Code	Course Title	No. of Periods/week	Total No. of periods	Marks for FA	Marks for SA
EE-102	Engineering Mathematics-I	5	150	20	80

Chapter No.	Title	No. of Periods	Weightage	No. of short question (3 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Algebra	31	22	4	1	CO1
2	Trigonometry	44	29	3	2	CO2
3	Co-ordinate Geometry	23	13	1	1	CO3
4	Differential Calculus	34	26	2	2	CO4
5	Applications of Derivatives	18	20	0	2	CO5
<b>TOTAL</b>		<b>150</b>	<b>110</b>	<b>10</b>	<b>8</b>	

S.No.	Chapter	No. of Periods	Marks Allotted	Short type	Essay type	CO's mapped
<b>Unit - I: Algebra</b>						
1	Functions	6	3	1	0	CO1
2	Partial Fractions	5	3	1	0	CO1
3	Matrices and Determinants	20	16	2	1	CO1
<b>Unit - II: Trigonometry</b>						
4	Trigonometric Ratios	2	0	0	0	CO2
5	Compound Angles	5	3	1	0	CO2
6	Multiple and Submultiple angles	8	3	1	0	CO2
7	Transformations	6	5	0	1/2	CO2
8	Inverse Trigonometric Functions	6	5	0	1/2	CO2
9	Trigonometric Equations	6	5	0	1/2	CO2
10	Properties of triangles	5	5	0	1/2	CO2
11	Complex Numbers	6	3	1	0	CO2
<b>Unit III: Co-ordinate Geometry</b>						
12	Straight Lines	5	3	1	0	CO3
13	Circles	6	5	0	1/2	CO3
14	Conic Sections	12	5	0	1/2	CO3
<b>Unit – IV: Differential Calculus</b>						
15	Limits and Continuity	6	3	1	0	CO4
16	Differentiation	28	23	1	2	CO4
<b>Unit – V: Applications of Derivatives</b>						

17	Geometrical Applications	4	5	0	1/2	CO5
18	Physical Applications	6	5	0	1/2	CO5
19	Maxima and Minima	4	5	0	1/2	CO5
20	Errors and Approximations	4	5	0	1/2	CO5
	Total	150	110	10	8	
			<b>Marks</b>	<b>30</b>	<b>80</b>	

<b>COURSE OBJECTIVES</b>	<p>(i) To apply the principles of Algebra, Trigonometry and Co-ordinate Geometry to real-time problems in engineering.</p> <p>(ii) To comprehend and apply the concept of Differential Calculus in engineering applications.</p>
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<b>COURSE OUTCOMES</b>	CO1	Identify functions as special relations, resolve partial fractions and solve problems on matrices and determinants.
	CO2	Solve problems using the concept of trigonometric functions, their inverses and complex numbers.
	CO3	Find the equations and properties of straight lines, circles and conic sections in coordinate system.
	CO4	Evaluate the limits and derivatives of various functions.
	CO5	Find solutions for engineering problems using differentiation.

## LEARNING OUTCOMES

### UNIT - I

**C.O. 1 Identify functions, resolve partial fractions and solve problems on matrices and determinants.**

**L.O.** 1.1 Define Set, Ordered pair and Cartesian product of two sets - examples.

1.2 Explain Relations and Functions – examples

1.3 Find Domain & Range of functions – simple examples.

1.4 Define one-one and onto functions.

1.5 Find the inverse of a function – simple examples.

1.6 Define rational, proper and improper fractions of polynomials.

1.7 Explain the procedure of resolving proper fractions of the types mentioned below into partial fractions

$$i) \frac{f(x)}{(ax+b)(cx+d)} \quad ii) \frac{f(x)}{(ax+b)^2(cx+d)}$$

1.8 Define a matrix and order of a matrix.

1.9 State various types of matrices with examples (emphasis on 3<sup>rd</sup> order square matrices).

1.10 Compute sum, difference, scalar multiplication and product of matrices. Illustrate the properties of these operations such as commutative, associative and distributive properties with examples and counter examples.

1.11 Define the transpose of a matrix and state its properties – examples.

- 1.12 Define symmetric and skew-symmetric matrices with examples. Resolve a square matrix into a sum of symmetric and skew-symmetric matrices and provide examples.
- 1.13 Define determinant of a square matrix; minor, co-factor of an element of a 3x3 square matrix with examples. Expand the determinant of a 3 x 3 matrix using Laplace expansion formula. State and apply the properties of determinants to solve problems.
- 1.14 Distinguish singular and non-singular matrices. Define multiplicative inverse of a matrix and list properties of adjoint and inverse. Compute adjoint and multiplicative inverse of a square matrix.
- 1.15 Solve a system of 3 linear equations in 3 unknowns using Cramer's rule and matrix inversion method.

## UNIT - II

### **C.O. 2 Solve problems using the concept of trigonometric functions, their inverses and complex numbers.**

- L.O.**
- 2.1 Define trigonometric ratios of any angle - List the values of trigonometric ratios at specified values.
  - 2.2 Draw graphs of trigonometric functions - Explain periodicity of trigonometric functions.
  - 2.3 Define compound angles and state the formulae of  $\sin(A\pm B)$ ,  $\cos(A\pm B)$ ,  $\tan(A\pm B)$  and  $\cot(A\pm B)$ .
  - 2.4 Give simple examples on compound angles to derive the values of  $\sin 15^\circ$ ,  $\cos 15^\circ$ ,  $\sin 75^\circ$ ,  $\cos 75^\circ$ ,  $\tan 15^\circ$ ,  $\tan 75^\circ$  etc.
  - 2.5 Derive identities like  $\sin(A+B) \sin(A-B) = \sin^2 A - \sin^2 B$  etc.
  - 2.6 Solve simple problems on compound angles.
  - 2.7 Derive the formulae of multiple angles  $2A$ ,  $3A$  etc and sub multiple angle  $A/2$  in terms of angle  $A$  of trigonometric functions.
  - 2.8 Derive useful allied formulae like  $\sin^2 A = (1 - \cos 2A)/2$  etc.
  - 2.9 Solve simple problems using the multiple and submultiple formulae.
- Syllabus for Unit test-I completed
- 2.10 Derive the formulae on transforming sum or difference of two trigonometric ratios in to a product and vice versa - examples on these formulae.
  - 2.11 Solve problems by applying these formulae to sum or difference or product of two terms.
  - 2.12 Explain the concept of the inverse of a trigonometric function by selecting an appropriate domain and range.
  - 2.13 Define inverses of six trigonometric functions along with their domains and ranges.
  - 2.14 Derive relations between inverse trigonometric functions so that the given inverse trigonometric function can be expressed in terms of other inverse trigonometric functions with examples.
  - 2.15 State various properties of inverse trigonometric functions and identities like  $\sin^{-1}x + \cos^{-1}x = \frac{\pi}{2}$ , etc.
  - 2.16 Apply formulae like  $\tan^{-1}x + \tan^{-1}y = \tan^{-1}\left(\frac{x+y}{1-xy}\right)$ , where  $x \geq 0, y \geq 0, xy < 1$  etc., to solve Simple problems.
  - 2.17 Explain what is meant by solution of trigonometric equations and find the general solutions of  $\sin x = k$ ,  $\cos x = k$  and  $\tan x = k$  with appropriate examples.
  - 2.18 Solve models of the type  $a \sin^2 x + b \sin x + c = 0$  and  $a \sin x + b \cos x = c$ .

- 2.19 State sine rule, cosine rule, tangent rule and projection rule and solve a triangle using these formulae.
- 2.20 List various formulae for the area of a triangle with examples.
- 2.21 Define a complex number, its modulus, conjugate, amplitude and list their properties.
- 2.22 Define arithmetic operations on complex numbers with examples.
- 2.23 Represent the complex number in various forms like modulus-amplitude (polar) form, Exponential (Euler) form with examples.

### UNIT - III

#### Coordinate Geometry

#### **C.O. 3 Find the equations and properties of straight lines, circles and conic sections in coordinate system.**

- L.O.** 3.1 Write different forms of a straight line – general form, point-slope form, slope-intercept form, two-point form, intercept form and normal form (or perpendicular form).
- 3.2 Find distance of a point from a line, acute angle between two lines, intersection of two non-parallel lines and distance between two parallel lines.
- 3.3 Define locus of a point and circle.
- 3.4 Write the general equation of a circle and find its centre and radius.
- 3.5 Find the equation of a circle, given (i) centre and radius, (ii) two ends of the diameter (iii) three non collinear points of type (0,0) (a,0), (0, b).
- 3.6 Define a conic section - Explain the terms focus, directrix, eccentricity, axes and latus-rectum of a conic with illustrations.
- 3.7 Find the equation of a conic when focus, directrix and eccentricity are given.
- 3.8 Describe the properties of Parabola, Ellipse and Hyperbola in standard forms whose axes are along the co-ordinate axes and solve simple examples on these conics.

Syllabus for Unit test-II completed

#### **C.O.4 Evaluate the limits and derivatives of various functions.**

- L.O.** 4.1 Explain the concept of limit and meaning of  $\lim_{x \rightarrow a} f(x) = l$  and state the properties of limits.

4.2 Evaluate the limits of the type  $\lim_{x \rightarrow a} \frac{f(x)}{g(x)}$  and  $\lim_{x \rightarrow \infty} \frac{f(x)}{g(x)}$

4.3 State the Standard limits  $\lim_{x \rightarrow a} \frac{x^n - a^n}{x - a}$ ,  $\lim_{x \rightarrow 0} \frac{\sin x}{x}$ ,  $\lim_{x \rightarrow 0} \frac{\tan x}{x}$ ,  $\lim_{x \rightarrow 0} \frac{a^x - 1}{x}$ ,

$\lim_{x \rightarrow 0} \frac{e^x - 1}{x}$ ,  $\lim_{x \rightarrow 0} (1 + x)^{\frac{1}{x}}$ ,  $\lim_{x \rightarrow \infty} \left(1 + \frac{1}{x}\right)^x$  (without proof) and solve simple problems using these standard limits.

- 4.4 Explain the concept of continuity of a function at a point and on an interval
- 4.5 State the concept of derivative of a function  $y = f(x)$  – definition, first principle as  $\lim_{h \rightarrow 0} \frac{f(x+h) - f(x)}{h}$  and also provide standard notations to denote the derivative of a function.

- 4.6 Explain the significance of derivative in scientific and engineering applications.
- 4.7 Find the derivative of standard algebraic, logarithmic, exponential and trigonometric functions using the first principle.
- 4.8 Find the derivatives of inverse trigonometric, hyperbolic and inverse hyperbolic functions.
- 4.9 State the rules of differentiation of sum, difference, scalar multiplication, product and quotient of functions with simple illustrative examples.
- 4.10 Explain the method of differentiation of a function of a function (Chain rule) with illustrative examples.
- 4.11 Explain the method of differentiation of parametric functions with examples.
- 4.12 Explain the procedure for finding the derivatives of implicit functions with examples.
- 4.13 Explain the need of taking logarithms for differentiating some functions of  $[f(x)]^{g(x)}$  type – examples on logarithmic differentiation.
- 4.14 Explain the concept of finding the second order derivatives with examples.
- 4.15 Explain the concept of functions of several variables, finding partial derivatives and difference between the ordinary and partial derivatives with simple examples.
- 4.16 Explain the concept of finding second order partial derivatives with simple problems.

**C.O. 5 Evaluate solutions for engineering problems using differentiation**

- L.O. 5.1 State the geometrical meaning of the derivative - Explain the concept of derivative to find the slopes of tangent and normal to a given curve at any point on it with examples.
- 5.2 Find the equations of tangent and normal to to a given curve at any point on it – simple problems.
- 5.3 Explain the derivative as a rate of change in distance-time relations to find the velocity and acceleration of a moving particle with examples.
- 5.4 Explain the derivative as a rate measurer in the problems where the quantities like areas, volumes vary with respect to time- illustrative examples.
  
- 5.5 Define the concept of increasing and decreasing functions - Explain the conditions to find points where the given function is increasing or decreasing with illustrative examples.
- 5.6 Explain the procedure to find the extreme values (maxima or minima) of a function of single variable- simple problems for quadratic and cubic polynomials.
- 5.7 Apply the concept of derivatives to find the errors and approximations - simple problems.

Syllabus for Unit test-III completed

**CO/PO – Mapping**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PSO1	PSO2	PSO3
CO1	3	2	2	3				3	2	2
CO2	3	3	2	2				3	2	2
CO3	3	3	2	2				3	2	2
CO4	3	3	3	3				3	3	3
CO5	3	3	3	3				3	3	3
<b>Avg.</b>	3	2.8	2.4	2.6				3	2.4	2.4

**3** = Strongly mapped (High), **2** = moderately mapped (Medium), **1** = slightly mapped (Low)

**Note:** The gaps in CO/PO mapping can be met with appropriate activities as follows:

For PO5: Appropriate quiz programmes may be conducted at intervals and duration as decided by concerned faculty.

For PO6: Seminars on applications of mathematics in various engineering disciplines are to be planned and conducted.

For PO7: Plan activities in such a way that students can visit the Library to refer standard books on Mathematics and access the latest updates in reputed national and international journals. Additionally, encourage them to attend seminars and learn mathematical software tools.

**PO- CO – Mapping strength**

PO No	Mapped with CO no	CO periods addressing PO in column I		Level (1,2 or 3)	Remarks
		No	%		
1	CO1, CO2, CO3, CO4, CO5	150 (31+44+23+34+18)	100%	3	>40% Level 3 Highly addressed 25% to 40% Level 2 Moderately addressed 5% to 25% Level 1 Low addressed <5% Not addressed
2	CO1, CO2, CO3, CO4, CO5	80 (8+23+12+22+15)	53.3%	3	
3	CO1, CO2, CO3, CO4, CO5	61 (9+14+9+14+15)	40.6%	3	
4	CO1, CO2, CO3, CO4, CO5	61 (14+9+9+14+15)	40.6%	3	
PSO 1	CO1, CO2, CO3, CO4, CO5	150 (31+44+23+34+18)	100%	3	
PSO 2	CO1, CO2, CO3, CO4, CO5	62 (10+14+9+14+15)	41.3%	3	
PSO 3	CO1, CO2, CO3, CO4, CO5	62 (10+14+9+14+15)	41.3%	3	

**COURSE CONTENT**

**Unit-I  
Algebra**

**1. Functions:**

Definitions of Set, Ordered pair, Cartesian product of two sets, Relations, Functions, Domain & Range of functions – One-one and onto functions, inverse of a function.

**2. Partial Fractions:**

Definitions of rational, proper and improper fractions of polynomials. Resolve rational fractions (proper fractions) into partial fractions covering the types mentioned below.

$$i) \frac{f(x)}{(ax+b)(cx+d)}$$

$$ii) \frac{f(x)}{(ax+b)^2(cx+d)}$$

### 3. Matrices:

Definition of a matrix, types of matrices - Algebra of matrices, equality of two matrices, sum, difference, scalar multiplication and product of matrices. Transpose of a matrix, Symmetric, skew-symmetric matrices - Determinant of a square matrix, minor and cofactor of an element, Laplace's expansion, properties of determinants - Singular and non-singular matrices, Adjoint and multiplicative inverse of a square matrix-System of linear equations in 3 variables-Solutions by Cramer's rule and Matrix inversion method.

## Unit-II

### Trigonometry

#### 4. Trigonometric ratios:

Definition of trigonometric ratios of any angle, values of trigonometric ratios at specified values, draw graphs of trigonometric functions, periodicity of trigonometric functions.

#### 5. Compound angles:

Formulas of  $\sin(A \pm B)$ ,  $\cos(A \pm B)$ ,  $\tan(A \pm B)$ ,  $\cot(A \pm B)$ , and related identities.

#### 6. Multiple and sub multiple angles:

Formulae for trigonometric ratios of multiple angles  $2A$ ,  $3A$  and sub multiple angle  $A/2$ .

#### 7. Transformations:

Transformations of products into sums or differences and vice versa.

#### 8. Inverse trigonometric functions:

Definition, domains and ranges-basic properties.

#### 9. Trigonometric equations:

Concept of a solution, principal value and general solution of trigonometric equations:

$\sin x = k$ ,  $\cos x = k$ ,  $\tan x = k$ , where  $k$  is a constant. Solutions of simple quadratic equations and equations of type  $a \sin x + b \cos x = c$ .

#### 10. Properties of triangles:

Relations between sides and angles of a triangle- sine rule, cosine rule, tangent rule and projection rule-area of a triangle.

#### 11. Complex Numbers:

Definition of a complex number, modulus, conjugate and amplitude of a complex number - Arithmetic operations on complex numbers - Modulus-Amplitude (polar) form, Exponential form (Euler form) of a complex number.

## UNIT-III

### Coordinate geometry

**12. Straight lines:** Various forms of a straight line - Angle between two lines, perpendicular distance from a point, intersection of non-parallel lines and distance between parallel lines.

**13. Circle:** Locus of a point, Circle definition - Circle equation given (i) centre and radius, (ii) two ends of a diameter (iii) three non-collinear points of type  $(0,0)$ ,  $(a,0)$ ,  $(0, b)$  - General equation of a circle -its centre and radius.

**14.** Definition of a conic section - Equation of a conic when focus, directrix and eccentricity are given - Properties of parabola, ellipse and hyperbola in standard forms.

## UNIT-IV

### Differential Calculus

**15. Concept of Limit-** Definition and Properties of Limits and Standard Limits - Continuity of a function at a point.

**16. Concept of derivative-** Definition (first principle)- different notations - Derivatives of standard algebraic, logarithmic, exponential, trigonometric, inverse trigonometric, hyperbolic and inverse

hyperbolic functions - Derivatives of sum, difference, scalar multiplication, product, quotient of functions - Chain rule, derivatives of parametric functions, derivatives of implicit functions, logarithmic differentiation - Second order derivatives - Functions of several variables, first and second order partial derivatives.

#### **UNIT-V**

##### **Applications of Derivatives**

17. Geometrical meaning of the derivative, equations of tangent and normal to a curve at any point.
18. Physical applications of derivatives – Velocity, acceleration, derivative as a rate measurer.
19. Applications of the derivative to find the extreme values – Increasing and decreasing functions, maxima and minima for quadratic and cubic polynomials.
20. Absolute, relative and percentage errors - Approximate values due to errors in measurements.

##### **Textbook:**

Engineering Mathematics-I, a textbook for first year diploma courses, prepared & prescribed by SBTET, AP.

##### **Reference Books:**

1. Shanti Narayan, A Textbook of matrices, S.Chand&Co.
2. Robert E. Moyer & Frank Ayers Jr., Schaum's Outline of Trigonometry, 4<sup>th</sup> Edition, Schaum's Series.
3. G.B.Thomas, R.L.Finney, Calculus and Analytic Geometry, Addison Wesley, 9<sup>th</sup> Edition, 1995.
4. Frank Ayers & Elliott Mendelson, Schaum's Outline of Calculus, Schaum's Series.
5. M.Vygodsky, Mathematical Handbook, Mir Publishers, Moscow.

### ENGINEERING PHYSICS

Course code	Course title	No.of periods per week	Total no. of periods	Marks for FA	Marks for SA
EE -103	Engineering Physics	03	90	20	80

Chapter No.	Title	No. of Periods	Weightage	No. of short question (3 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Units and measurements	09	03	1	0	CO1
2	Statics	11	13	1	1	
3	Gravitation	12	20	0	2	CO2
4	Concepts of energy	10	13	1	1	
5	Thermal physics	10	13	1	1	CO3
6	Sound	12	16	2	1	
7	Electricity & Magnetism	13	16	2	1	CO4
8	Modern physics	13	16	2	1	
<b>TOTAL</b>		<b>90</b>	<b>110</b>	<b>10</b>	<b>8</b>	

<b>COURSE OBJECTIVES</b>	<p>(1) To understand the basic concepts of physics for various Engineering applications as required for industries.</p> <p>(2) To equip the students with the scientific advances in technology and make the student suitable for any industrial or scientific organization.</p>
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<b>COURSE OUTCOMES</b>	CO1	Familiarize with various physical quantities, their SI units and errors in measurements; understand the concepts of vectors and various forces in statics.
	CO2	Understand the concepts of gravitation with reference to applications in satellites, provide the knowledge of various forms of energy and their working principles.
	CO3	Familiarize with the knowledge of transmission of heat and gas laws; provide the knowledge on musical sound and noise as pollution and also the concepts of echo and reverberation.
	CO4	Provide basic knowledge of electricity and concepts of magnetism and magnetic materials; familiarize with the advances in Physics such as photoelectric cell, optical fibers, semiconductors, superconductors and nanotechnology.

**Matrix showing mapping of Course Outcomes with Program Outcomes**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PSO1	PSO2	PSO3
CO1	3	2	2	1			2	2		2
CO2	3	2	2	2	2		2	1		2
CO3	2		1		2		1		1	1
CO4	3	2	3	2	2		3	2		2

**CO-PO Mapping Strength**

<b>Course code EE-103</b>	<b>Engineering Physics No of Course Objectives : 4</b>				<b>No of periods 90</b>
POs	Mapped with CO No	CO periods addressing PO in Col 1 NO	%	1,2,3	remarks
PO1	CO1,CO2,CO3,CO4	44	48.9 %	3	>40% level 3 (highly addressed)  25% to 40% level 2 (moderately addressed)  5% to 25% level 1 (Low addressed)  < 5% (not addressed)
PO2	CO1,CO2, CO4	11	12.2%	1	
PO3	CO1, CO2,CO3, CO4	10	11.1%	1	
PO4	CO1, CO2,CO4	8	8.9%	1	
PO5	CO2,CO3, CO4	8	8.9%	1	
PO6					
PO7	CO1, CO2, CO3, CO4	9	10.0%	1	

3 = strongly mapped, 2 = moderately mapped, 1 = slightly mapped

**Note: The gaps in CO and PO mapping will be achieved by one or more appropriate activities from the following.**

- |                        |                        |                                |                     |
|------------------------|------------------------|--------------------------------|---------------------|
| (i) Seminars           | (ii) Tutorials         | (iii) Guest Lecturers          | (iv) Assignments    |
| (v) Quiz competitions  | (vi) Industrial visits | (vii) Tech fest                | (viii) Mini project |
| (ix) Group discussions | (x) Virtual classes    | (xi) Library visit for e-books |                     |

## LEARNING OUTCOMES

Upon completion of the course the student shall be able to

### 1.0 Understand the concept of units and measurements

- 1.1 Explain the concept of units
- 1.2 Define the terms
  - a) Physical quantity, b) Fundamental physical quantities and
  - c) Derived physical quantities
- 1.3 Define unit
- 1.4 Define fundamental units and derived units
- 1.5 State SI units with symbols for fundamental and some derived quantities
- 1.6 State Multiples and Submultiples in SI system
- 1.7 State rules of writing S.I units
- 1.8 State advantages of SI units
- 1.9 What are direct and indirect measurements.
- 1.10 Define accuracy and least count
- 1.11 Define error in measurement
- 1.12 Define absolute, relative and percentage errors with their formulae
- 1.13 Solve simple problems on absolute, relative and percentage errors

### 2.0 Understand the concepts of statics

- 2.1 Explain the concept of Vectors
- 2.2 Define scalar and vector quantities with examples
- 2.3 Represent vectors geometrically
- 2.4 Define the types of vectors (equal, negative, unit, co-initial, co-planar, position vector)
- 2.5 Resolve the vector into rectangular components
- 2.6 State and explain triangle law of addition of vectors
- 2.7 Define concurrent forces, co-planar forces and equilibrant.
- 2.8 State and explain Lami's theorem
- 2.9 State the parallelogram law of addition of forces with diagram.
- 2.10 Write the expressions for magnitude and direction of resultant (no derivation)
- 2.11 Illustrate parallelogram law with examples (i) flying of bird and (ii) working of sling.
- 2.12 Define moment of force and couple.
- 2.13 Write the formulae and S.I units of moment of force and couple.
- 2.14 Solve simple problems on (i) Resolution of force and (ii) Parallelogram law of forces (finding  $R, \alpha$  and  $\theta$ ).

### 3.0 Understand the concepts of Gravitation

- 3.1 State and explain Newton's universal law of gravitation.
- 3.2 Define  $G$  and mention its value.
- 3.3 Explain the acceleration due to gravity ( $g$ )
- 3.4 Explain the factors affecting the value of  $g$
- 3.5 Derive the relationship between  $g$  and  $G$ .
- 3.6. State and explain the Kepler's laws of planetary motion
- 3.7 Define a satellite.
- 3.8 What are natural and artificial satellites? Give examples.
- 3.9 Define orbital velocity and write its formula.
- 3.10 Define escape velocity and write its formula.
- 3.11 Write a brief note on Polar satellites.
- 3.12 Write a brief note on Geo-stationary satellites.
- 3.13 Mention the applications of artificial satellites.

3.14 Solve simple problems on (i) Newton's law of gravitation and (ii) calculation of orbital and escape velocities.

**4.0 Understand the concepts of Energy.**

4.1 Define work done and energy. Mention their SI units.

4.2 List various types of energy.

4.3 Define P.E with examples. Write its equation.

4.4 Define K.E with examples. Write its equation.

4.5 Derive relationship between K.E and momentum.

4.6 State the law of conservation of energy. Give various examples.

4.7 Write a brief note on solar energy.

4.8 Explain the principle of solar thermal conversion.

4.9 Explain the principle of photo voltaic effect

4.10 Solve simple problems on (i) work done (ii) P.E & K.E and (iii) Relation between K.E & momentum.

**5.0 Understand the concepts of thermal physics**

5.1 Define the concepts of heat and temperature

5.2 State different modes of transmission of heat

5.3 Explain conduction, convection and radiation with two examples each.

5.4 State and explain Boyle's law

5.5 Define absolute zero temperature

5.6 Explain absolute scale of temperature

5.7 State the relationship between degree Celsius, Kelvin and Fahrenheit temperatures

5.8 State Charle's law and write its equation

5.9 State Gay-Lussac's law and write its equation

5.10 Define ideal gas

5.11 Derive ideal gas equation

5.12 Explain why universal gas constant (R) is same for all gases

5.13 Calculate the value of R for 1 gram mole of gas.

5.14 Solve simple problems on (i) Inter conversion of temperatures between °C, K and F (ii) Gas laws and (iii) Ideal gas equation.

**6.0 Understand the concepts of Sound**

6.1 Define the term sound

6.2 Define longitudinal and transverse waves with one example each

6.3 Explain the factors which affect the velocity of sound in air

6.4 Distinguish between musical sound and noise

6.5 Explain noise pollution and state SI unit for intensity of sound

6.6 Explain sources of noise pollution

6.7 Explain effects of noise pollution

6.8 Explain methods of minimizing noise pollution

6.9 Define Doppler effect.

6.10 List the Applications of Doppler effect

6.11 Define reverberation and reverberation time

6.12 Write Sabine's formula and name the physical quantities in it.

6.13 Define echoes and explain the condition to hear an echo.

6.14 Mention the methods of reducing an echo

6.15 Mention the applications of an echo

6.16 What are ultra sonics

6.17 Mention the applications of ultra sonics, SONAR

6.18 Solve simple problems on echo

**7.0 Understand the concepts of Electricity and Magnetism**

- 7.1 Explain the concept of P.D and EMF
- 7.2 State Ohm's law and write the formula
- 7.3 Explain Ohm's law
- 7.4 Define resistance and specific resistance. Write their S.I units.
- 7.5 State and explain Kichoff's first law.
- 7.6 State and explain Kirchoff's second law.
- 7.7 Describe Wheatstone bridge with legible sketch.
- 7.8 Derive an expression for balancing condition of Wheatstone bridge.
- 7.9 Describe Meter Bridge experiment with necessary circuit diagram.
- 7.10 Write the formulae to find resistance and specific resistance in meter bridge
- 7.11 Explain the concept of magnetism
- 7.12 What are natural and artificial magnets (mention some types)
- 7.13 Define magnetic field and magnetic lines of force.
- 7.14 Write the properties of magnetic lines of force
- 7.15 State and explain the Coulomb's inverse square law of magnetism
- 7.16 Define magnetic permeability
- 7.17 Define para, dia, ferro magnetic materials with examples
- 7.18 Solve simple problems on (i) Ohm's law (ii) Kirchoff's first law (iii) Wheatstone bridge (iv) meter bridge and (v) Coulomb's inverse square law
- 8.0 Understand the concepts of Modern physics**
- 8.1 State and explain Photo-electric effect.
- 8.2 Write Einstein's Photo electric equation and name the physical quantities in it.
- 8.3 State laws of photo electric effect
- 8.4 Explain the Working of photo electric cell
- 8.5 List the Applications of photoelectric effect
- 8.6 Recapitulate refraction of light and its laws
- 8.7 Define critical angle
- 8.8 Explain the Total Internal Reflection
- 8.9 Explain the principle and working of Optical Fiber
- 8.10 List the applications of Optical Fiber
- 8.11 Explain the energy gap based on band structure
- 8.12 Distinguish between conductors, semiconductors and insulators based on energy gap
- 8.13 Define doping
- 8.14 Explain the concept of hole
- 8.15 Explain the types of semiconductors : Intrinsic and extrinsic
- 8.16 Explain n-type and p-type semiconductors
- 8.17 Mention the applications of semiconductors
- 8.18 Define superconductor and superconductivity
- 8.19 List the applications of superconductors
- 8.20 Nanotechnology definition, nano materials and applications

## **COURSECONTENT**

### **1. Units and measurements**

Introduction – Physical quantity – Fundamental and Derived quantities – Fundamental and derived units - SI units – Multiples and Sub multiples – Rules for writing S.I. units-Advantages of SI units – Direct and indirect measurements – Accuracy and least count – Errors : Absolute, relative and percentage errors –Problems.

## 2. Statics

Scalars and Vectors– Representation of a vector - Types of vectors - Resolution of vector into rectangular components – Triangle law of vectors – Concurrent forces - Lami's theorem - Parallelogram law of forces : Statement, equations for magnitude and direction of resultant, examples – Moment of force and couple – Problems.

## 3. Gravitation

Newton's law of gravitation and  $G$  – Concept of acceleration due to gravity ( $g$ ) – Factors affecting the value of  $g$  – Relation between  $g$  and  $G$ - Kepler's laws – Satellites : Natural and artificial – Orbital velocity and escape velocity – Polar and geostationary satellites – Applications of artificial satellites – Problems.

## 4. Concepts of energy

Work done & Energy-Definition and types of energy - potential energy - kinetic energy-- K.E and momentum relation – Law of Conservation of energy, examples - Solar energy, principles of thermal and photo conversion – Problems.

## 5. Thermal physics

Modes of transmission of heat – Expansion of Gases - Boyle's law – Absolute scale of temperature - Thermometric scales and their inter conversion - Charle's law - Gay-Lussac's law - Ideal gas equation - Universal gas constant ( $R$ ) - Problems.

## 6. Sound

Sound - Nature of sound - Types of wave motion, Longitudinal and transverse – Factors affecting the velocity of sound in air - musical sound and noise - Noise pollution – Causes & effects - Methods of reducing noise pollution - Doppler effect - Echo- Reverberation -Reverberation time-Sabine 's formula – Ultrasonics & applications – SONAR - Problems.

## 7. Electricity & Magnetism

Concept of P.D and EMF - Ohm's law and explanation-Specific resistance - Kirchoff's laws – Wheat stone's bridge - Meter bridge.  
Natural and artificial magnets – magnetic field and magnetic lines of force – Coulomb's inverse square law – Permeability – Magnetic materials – Para, dia, ferro – Examples – Problems.

## 8. Modern Physics

Photoelectric effect – laws of photoelectric effect – photoelectric cell – Applications of photoelectric cell - Total internal reflection - Fiber optics - Principle and working of an optical fiber - Applications of optical fibers – Semiconductors : Based on Energy gap – Doping – Hole - Intrinsic and extrinsic semiconductors (n-type & p-type) – Applications of semiconductors – Superconductivity – applications – Nanotechnology definition, nano materials, applications.

## REFERENCES

- |  |                                   |
|--|-----------------------------------|
| 1. Intermediate physics - Volume - I & 2                   | Telugu Academy (English version)  |
| 2. Unified physics Volume 1, 2, 3 and 4                    | Dr. S.L Guptha and Sanjeev Guptha |
| 3. Concepts of Physics, Vol 1 & 2                          | H.C. Verma                        |
| 4. Text book of physics Volume I & 2                       | Resnick & Halliday                |
| 5. Fundamentals of physics                                 | Brijlal & Subramanyam             |
| 6. Text book of applied physics                            | Dhanpath Roy                      |
| 7. NCERT Text Books of physics                             | Class XI & XII Standard           |
| 8. e-books/e-tools/websites/Learning Physics software/eLMS |                                   |

**Table showing the scope of syllabus to be covered for unit tests**

<b>Unit test</b>	<b>Learning outcomes to be covered</b>
Unit test - 1	From 1.1 to 3.14
Unit test - 2	From 4.1 to 6.18
Unit test - 3	From 7.1 to 8.20

## ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Course code	Course Title	No. of Periods per week	Total No. of Periods	Marks for FA	Marks for SA
EE-104	Engineering Chemistry and Environmental Studies	3	90	20	80

Chapter No.	Title	No. of Periods	Weightage	No. of short question (3 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Fundamentals of Chemistry	14	21	2	1.5	CO1
2	Solutions, Acids and Bases	16	21	2	1.5	CO1
3	Electrochemistry	12	13	1	1	CO2
4	Corrosion	8	13	1	1	CO2
5	Water Treatment	8	13	1	1	CO3
6	Polymers & Engineering Materials	12	13	1	1	CO4
7	Fuels	6	3	1	0	CO4
8	Environmental Studies	14	13	1	1	CO5
<b>TOTAL</b>		<b>90</b>	<b>110</b>	<b>10</b>	<b>8</b>	

<b>COURSE OBJECTIVES</b>	<ol style="list-style-type: none"> <li>1. To familiarize with the concepts of chemistry involved in the process of various Engineering Industrial Applications.</li> <li>2. To know the various natural and man-made environmental issues and concerns with an interdisciplinary approach that include physical, chemical, biological and socio cultural aspects of environment.</li> <li>3. to reinforce theoretical concepts by conducting relevant experiments/exercises</li> </ol>
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<b>COURSE OUTCOMES</b>	CO1	Explain Bohr's atomic model, chemical bonding, mole concept, acids and bases, P <sup>H</sup> and Buffer solutions.
	CO2	Explain electrolysis, Galvanic cell, batteries and corrosion
	CO3	Explain the chemistry involved in the treatment of hardness in water.
	CO4	Explain the methods of preparation and applications of Polymers and Elastomers, chemical composition and applications of Alloys, Composite Materials, Liquid Crystals, Nano Materials and Fuels.
	CO5	Explain Global impacts due to air pollution, causes, effects and controlling methods of water pollution and understand the environment, forest resources, e-Pollution and Green Chemistry Principles.

EE-104	Engineering. Chemistry and Environmental studies No of Course Outcomes:5				No Of periods 90
POs	Mapped with CO No	CO periods addressing PO in Col NO. 1	%	Level 1,2,3	remarks
PO1	CO1,CO2,CO3	42	46.7 %	3	>40% level 3 (highly addressed) 25% to 40% level 2(moderately addressed) 5% to 25% level 1 (Low addressed) < 5%(not addressed)
PO2	CO2,CO3	16	17.8%	1	
PO3	CO4	12	13.3%	1	
PO4	CO4	6	6.7%	1	
PO5	CO5	14	15.5%	1	
PO6					
PO7					

**COs-POs mapping strength (as per given table)**

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PSO1	PSO2	PSO3
CO1	3	-	-	-	-	-	-	-	-	-
CO2	3	1	-	-	-	-	-	-	-	-
CO3	3	1	-	-	-	-	-	-	-	-
CO4	-	-	1	1	-	-	1	-	-	-
CO5	-	-	-	-	1	-	-	1	-	-
Average	3	1	1	1	1		1	1	1	-

3=strongly mapped 2= moderately mapped 1= slightly mapped

Note: The gaps in CO and PO mapping will be achieved by one or more appropriate activities from the following:

- i) Seminars ii) Tutorials iii) Guest Lectures iv) Assignments v) Quiz competitions vi) Industrial visit vii) Tech Fest viii) Mini project ix) Group discussions x) Virtual classes xi) Library visit for e-books

## LEARNING OUTCOMES

### 1.0 Atomic structure

- 1.1 Explain the charge, mass of fundamental particles of an atom (electron, proton and neutron) and the concept of atomic number and mass number.
- 1.2 State the Postulates of Bohr's atomic theory and its limitations.
- 1.3 Explain the significance of four Quantum numbers and draw the atomic structures of Silicon and Germanium.
- 1.4 Define Orbital of an atom and draw the shapes of s,p and d-orbitals.
- 1.5 Explain 1. Aufbau principle, 2. Pauli's exclusion principle 3. Hund's principle.
- 1.6 Write the electronic configuration of elements up to atomic number 30.
- 1.7 Explain the significance of chemical bonding.
- 1.8 Explain the Postulates of Electronic theory of valency.
- 1.9 Define and explain Ionic and Covalent bonds with examples of NaCl , \*H<sub>2</sub>, \*O<sub>2</sub> and \*N<sub>2</sub>.(\* Lewis dot method).
- 1.10 List out the Properties of Ionic compounds and covalent compounds and distinguish between their properties.

### 2.0 Solutions, Acids and Bases

- 2.1 Define the terms 1. Solution, 2. Solute and 3. Solvent.
- 2.2 Classify solutions based on solubility.
- 2.3 Define the terms 1. Atomic weight, 2. Molecular weight and 3. Equivalent weight. Calculate Molecular weight and Equivalent weight of the given acids (HCl, H<sub>2</sub>SO<sub>4</sub>, H<sub>3</sub>PO<sub>4</sub>) Bases (NaOH, Ca(OH)<sub>2</sub>, Al(OH)<sub>3</sub> and Salts (NaCl, Na<sub>2</sub>CO<sub>3</sub>, CaCO<sub>3</sub>).
- 2.4 Define mole and solve numerical problems on mole concept.
- 2.5 Define molarity, normality and solve numerical problems on molarity and normality.
  - a) Calculate the Molarity or Normality, if weight of solute and volume of solution are given.
  - b) Calculate the weight of solute, if Molarity or Normality with volume of solution are given.
  - c) Problems on dilution to convert high concentrated solutions to low concentrated solutions.
- 2.6 Explain Arrhenius theory of Acids and Bases and give its limitations.
- 2.7 Define ionic product of water, pH and solve numerical problems on pH (Strong Acids and Bases).
- 2.8 Define buffer solution and classify buffer solutions with examples. Give its applications.

### 3.0 Electrochemistry

- 3.1 Define the terms 1. Conductor 2. Semiconductor 3. Insulator, 4. Electrolyte 5. Non-electrolyte. Give two examples each.
- 3.2 Distinguish between Metallic conduction and Electrolytic conduction.
- 3.3 Explain electrolysis by taking an example of used NaCl and list out the applications of electrolysis.
- 3.4 Define Galvanic cell. Explain the construction and working of Galvanic cell.
- 3.5 Distinguish between electrolytic cell and galvanic cell.
- 3.6 Define battery and list the types of batteries with examples.
- 3.7 Explain the construction, working and applications of i) Dry cell (Leclanche cell), ii) Lead storage battery, iii) Lithium-Ion battery and iv) Hydrogen-Oxygen fuel cell.

### 4.0 Corrosion

- 4.1 Define the term corrosion.
- 4.2 state the Factors influencing the rate of corrosion.
- 4.3 Describe the formation of (a) composition cell (b) stress cell (c) concentration cell during corrosion.
- 4.4 Define rusting of iron and explain the mechanism of rusting of iron.
- 4.5 Explain the methods of prevention of corrosion by
  - (a) Protective coatings (anodic and cathodic coatings).
  - (b) Cathodic protection (Sacrificial anode process and Impressed-voltage process).

## **5.0 Water Treatment**

- 5.1 Define soft water and hard water with respect to soap action.
- 5.2 Define and classify the hardness of water.
- 5.3 List out the salts that causing hardness of water (with Formulae).
- 5.4 State the disadvantages of using hard water in industries.
- 5.5 Define Degree of hardness and units of hardness (mg/L and ppm).
- 5.6 Solve numerical problems on hardness.
- 5.7 Explain the methods of softening of hard water by (i) Ion-exchange process and (ii) Reverse Osmosis process.

## **6.0 Polymers & Engineering materials.**

### **A) Polymers**

- 6.1 Explain the concept of polymerization.
- 6.2 Describe the methods of polymerization (a) addition polymerization of ethylene (b) condensation polymerization of Bakelite (Only flowchart).
- 6.3 Define plastic. Explain a method of preparation and uses of the following plastics:  
1. PVC 2. Teflon 3. Polystyrene 4. Nylon 6,6.
- 6.4 Define elastomers. Explain a method of preparation and applications of the following:  
1. Buna-S 2. Neoprene.

### **B) Engineering Materials**

- 6.5 Define an alloy. Write the composition and applications of the following:  
1. Nichrome 2. Duralumin 3. Stainless Steel.
- 6.6 Define Composite Materials and give any two examples. State their Properties and applications.
- 6.7 Define Liquid Crystals and give any two examples. State their Properties and applications.
- 6.8 Define Nano Materials and give any two examples. State their Properties and applications.

## **7.0 Fuels**

- 7.1 Define the term fuel.
- 7.2 Classify the fuels based on occurrence.
- 7.3 Write the composition and uses of the following:  
1. LPG 2. CNG 3. Biogas 4. Power alcohol
- 7.4 Write the commercial production of Hydrogen as future fuel. Give its advantages and disadvantages.

## **8.0 ENVIRONMENTAL STUDIES**

- 8.1 Explain the scope and importance of environmental studies.
- 8.2 Define environment. Explain the different segments of environment.  
1. Lithosphere 2. Hydrosphere 3. Atmosphere 4. Biosphere
- 8.3 Define the following terms:  
1. Pollutant 2. Pollution 3. Contaminant 4. Receptor 5. Sink 6. Particulates 7. Dissolved oxygen (DO) 8. Threshold Limit Value (TLV) 9. BOD 10. COD 11. Eco system 12. Producers 13. Consumers 14. Decomposers with examples.
- 8.4 State the renewable and non-renewable energy sources with examples.
- 8.5 State the uses of forest resources.
- 8.6 Explain the causes and effects of deforestation.
- 8.7 Define air pollution and explain its Global impacts 1. Greenhouse effect, 2. Ozone layer depletion and 3. Acid rain.
- 8.8 Define Water pollution. Explain the causes, effects and controlling methods of Water pollution.
- 8.9 Define e-Pollution, State the sources of e-waste. Explain its health effects and control methods.
- 8.10 Define Green Chemistry. Write the Principles and benefits of Green Chemistry.

## COURSE CONTENT

### ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

#### 1. Fundamentals of Chemistry

**Atomic Structure:** Introduction - Fundamental particles – Bohr's theory – Quantum numbers – Atomic structure of Silicon and Germanium - Orbitals, shapes of s, p and d orbitals - Aufbau's principle - Hund's rule - Pauli's exclusion Principle -Electronic configuration of elements.

**Chemical Bonding:** significance–Electronic theory of valency- Types of chemical bonds – Ionic and covalent bond with examples–Properties of Ionic and Covalent compounds.

#### 2. Solutions, Acids and Bases

**Solutions:** Types of solutions - Mole concept – Numerical problems on mole concept -Methods of expressing concentration of a solution – Molarity and Normality – Numerical problems on molarity and normality.

**Acids and Bases:** Arrhenius theory of acids and bases – Ionic product of water- pH–Numerical problems on pH–Buffer solutions – Classification- applications.

#### 3. Electrochemistry

Conductors, semiconductors, insulators, electrolytes and non-electrolytes – Electrolysis of fused NaCl–Applications of electrolysis - Galvanic cell – Battery-Types- Dry Cell (Leclanche Cell), Lead- Storage battery- Lithium-Ion battery -Hydrogen-Oxygen Fuel cell.

#### 4. Corrosion

Introduction - Factors influencing corrosion - Composition, Stress and Concentration Cells– Rusting of iron and its mechanism – Prevention of corrosion by Protective Coating methods, Cathodic Protection methods.

#### 5. Water treatment

Introduction– Soft and Hard water– Causes of hardness– Types of hardness– Disadvantages of hard water – Degree of hardness (ppm and mg/lit) – Numerical problems on hardness - Softening methods – Ion-Exchange process– Reverse Osmosis process.

#### 6. Polymers & Engineering materials

##### Polymers:

Concept of polymerization – Types of polymerization – Addition, condensation with examples – Plastics - Preparation and uses of i).PVC ii) Teflon iii) Polystyrene and iv) Nylon 6,6.

Elastomers: Preparation and application of i)Buna-s and ii) Neoprene.

##### Engineering Materials:

Alloys- Composition and applications of i) Nichrome, ii) Duralumin and iii) Stainless Steel.

Composite Materials- Properties and applications.

Liquid Crystals- Properties and applications.

Nano Materials- Properties and applications.

#### 7. Fuels

Definition and classification of fuels – Composition and uses of i) LPG ii) CNG iii) Biogas and iv) Power alcohol – Hydrogen as a future fuel-production- advantages and disadvantages.

#### 8. ENVIRONMENTAL STUDIES

Scope and importance of environmental studies – Environment - Important terms related to environment–Renewable and non-renewable energy sources–Forest resources – Deforestation – Air pollution–Global impacts on environment –Water pollution – causes – effects – control measures- e-Pollution –Sources of e-waste - Health effects - Control methods - Green Chemistry- Principles -Benefits.

## REFERENCE BOOKS

- |                           |                                 |
|---------------------------|---------------------------------|
| 1. Telugu Academy         | Intermediate chemistry Vol. 1&2 |
| 2. Jain & Jain            | Engineering Chemistry           |
| 3. O.P. Agarwal, Hi-Tech. | Engineering Chemistry           |
| 4. D.K.Sharma             | Engineering Chemistry           |
| 5. A.K. De                | Engineering Chemistry           |

## Syllabus for Unit tests

<b>Unit Test</b>	<b>Learning outcomes to be covered</b>
Unit Test – 1	<b>From 1.1 to 2.8</b>
Unit Test – 2	<b>From 3.1 to 5.7</b>
Unit Test – 3	<b>From 6.1 to 8.10</b>

### ELECTRICAL ENGINEERING MATERIAL SCIENCE

Course code	Course title	No. of periods/week	Total no. of periods	Marks for FA	Marks for SA
EE-105	ELECTRICAL ENGINEERING MATERIAL SCIENCE	04	120	20	80

Chapter No.	Title	No. of Periods	Weightage	No. of short question (3 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Conducting Materials and Semiconducting Materials	25	26	2	2	CO1
2	Insulating Materials and Di- electric Materials	30	26	2	2	CO2
3	Magnetic Materials	15	13	1	1	CO3
4	Magnetic effects of Electric Current and Electromagnetic Induction	30	29	3	2	CO4
5	Electrostatics and Capacitance	20	16	2	1	CO5
<b>TOTAL</b>		<b>120</b>	<b>110</b>	<b>10</b>	<b>8</b>	

<b>COURSE OBJECTIVES</b>	<ol style="list-style-type: none"> <li>1) To familiarise with the knowledge of different electrical engineering materials.</li> <li>2) To comprehend the magnetic effects of electric current.</li> <li>3) To know the concept of electro-magnetic induction and electrostatic field.</li> </ol>
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<b>COURSE OUTCOMES</b>	CO1	EE-105.1	Explain the properties of different conducting and semi-conducting materials and their applications
	CO2	EE-105.2	Analyze Insulating materials and dielectric materials
	CO3	EE-105.3	Describe Magnetic materials and their properties
	CO4	EE-105.4	Comprehend the principles of Magnetic effects of Electric Current and Electromagnetic Induction
	CO5	EE-105.5	Understand the concepts of Electrostatics and Capacitance

## LEARNING OUTCOMES

### 1. Conducting Materials and Semiconducting Materials

- 1.1 Define Conducting Materials
- 1.2 State the properties of conducting materials
- 1.3 Define the terms (i) Hardening (ii) Annealing
- 1.4 State the main requirements of (i) Low Resistivity Materials (ii) High resistivity materials
- 1.5 List some examples of i) Low Resistivity Materials ii) High Resistivity materials
- 1.6 Mention the Properties & Applications of Copper and Aluminium
- 1.7 Distinguish between Copper and Aluminium
- 1.8 Mention the properties & applications of ACSR Conductors and AAAC.
- 1.9 List the properties & Applications of High Resistive Materials of (i) Nichrome (ii) Tungsten (iii) Carbon
- 1.10 Define Semiconducting materials
- 1.11 Classify Semi conducting materials
- 1.12 Define (i) Intrinsic Semiconductors and (ii) Extrinsic Semiconductors
- 1.13 Distinguish between Intrinsic and Extrinsic semiconductors.
- 1.14 Explain the formation of (i) P – type semiconductor and (ii) N – type semiconductor
- 1.15 Distinguish between P and N type Semiconductors

### 2. Insulating Materials and Di-electric materials

- 2.1 Define Insulating Materials
- 2.2 Draw energy level diagrams of conductors, insulators and semi-Conductors
- 2.3 Distinguish between Conductors, Insulators and Semiconductors
- 2.4 Define Insulation resistance and explain factors affecting insulation resistance
- 2.5 Classify insulating materials
- 2.6 State the properties and applications of (i) Impregnated paper (ii) Wood (iii) Asbestos (iv) Mica (v) Ceramics (vi) Glass
- 2.7 Explain the properties and applications of PVC
- 2.8 State the effects of the following on P.V.C.  
(i) Filler (ii) Stabilizer (iii) Plasticizer (iv) Additives.
- 2.9 Know the Permittivity of commonly used di – electric materials  
(i) Air (ii) Bakelite (iii) Glass (iv) Mica (v) Paper (vi) Porcelain (vii) Transformer oil
- 2.10 Explain Polarization
- 2.11 Explain Di-electric Loss

2.12 List any four applications of Di-electrics

### 3. Magnetic Materials

- 3.1 Classify the Magnetic Materials  
(i) Ferro (ii) Para (iii) Dia-Magnetic materials with examples
- 3.2 Explain (i) Soft Magnetic materials (ii) Hard Magnetic materials
- 3.3 Draw (i) B-H Curve (ii) Hysteresis loop
- 3.4 Explain Hysteresis loop
- 3.5 Explain Hysteresis loss and State Steinmetz equation (No-Problems)
- 3.6 Explain Eddy Current Losses
- 3.7 State Curie point
- 3.8 Define Magnetostriction

### 4. Magnetic effects of Electric Current and Electromagnetic Induction

- 4.1 State Coulombs laws of Magnetism
- 4.2 Define the terms Absolute and Relative Permeability of medium and give relation between them
- 4.3 Explain the concept of lines of force & magnetic field
- 4.4 State Right hand Thumb rule
- 4.5 Draw the field patterns due to  
(i) Straight current carrying conductor (ii) Solenoid
- 4.6 State and list the applications of (i) Work law (ii) Biot- Savart's Law (Laplace law)
- 4.7 Explain the Mechanical force on a current carrying Conductor placed inside a Magnetic field.
- 4.8 Derive an expression for the magnitude of the force on a current carrying conductor inside a magnetic field.
- 4.9 State Fleming's Left Hand rule
- 4.10 Understand the concept of the Magnetic circuit and Define the terms MMF, Flux and Reluctance
- 4.11 Compare Magnetic circuit with Electric circuit in different aspects
- 4.12 Explain the effect of air gap in a magnetic circuit
- 4.13 Explain the terms leakage flux and leakage co-efficient
- 4.14 State Faraday's laws of Electro-Magnetic Induction
- 4.15 Explain Dynamically and Statically induced E.M.Fs
- 4.16 State Lenz's law
- 4.17 Explain Fleming's Right Hand rule
- 4.18 State the concept of Self and Mutual inductance and write their expressions
- 4.19 State Co-efficient of coupling
- 4.20 Develop an expression for the energy stored in a magnetic field

### 5. Electrostatics and Capacitance

- 5.1 State Coulomb's laws of Electrostatics and solve the problems
- 5.2 Define the following terms  
(i) Unit Charge (ii) Absolute permittivity (iii) Relative permittivity (iv) Electric Flux  
(v) Flux Density (vi) Field intensity
- 5.3 Draw the field patterns due to  
i) Isolated positive charge

- ii) Isolated negative charge
  - iii) Unlike charges placed side by side
  - iv) Like charges placed side by side
- 5.4 Compare Electrostatic and Magnetic lines of force in different aspects.
  - 5.5 Define the concept of electric potential and potential difference
  - 5.6 Define Di-electric strength and Di-electric constant
  - 5.7 Give the permittivity of commonly used Di-electric materials
  - 5.8 Define Capacitance and state factors affecting the capacitance of a capacitor
  - 5.9 Derive the formula for capacitance of a parallel plate capacitor
  - 5.10 State different types of capacitors with its uses
  - 5.11 Derive an expression for equivalent capacitance
    - i) When two Capacitors are connected in series
    - ii) When two Capacitors are connected in parallel
  - 5.12 Derive an expression for the Energy stored in a capacitor

## **HYPONATED COURSE CONTENTS**

### **1. Conducting Materials and Semiconducting materials:**

Conducting Materials – Properties -Hardening, Annealing - Low Resistivity Materials – Requirements – Properties and applications of Copper and Aluminium - Comparison between Copper and Aluminium - ACSR Conductors, AAAC - High Resistive Materials – Requirements- Properties and applications of Nichrome, Tungsten and Carbon-Semiconducting materials-classification-comparison between intrinsic and extrinsic semiconductors –Formation of P-type and N-type semiconductors-comparison of P-type and N-type semiconductors.

### **2. Insulating Materials and Di-electric materials**

Insulating materials – Energy level diagrams of Conductors, Insulators and Semiconductors - Distinguish between Conductors, Insulators and Semiconductors-Factors effecting Insulation resistance - Classification of Insulating materials - Properties & Applications of Impregnated Paper, Wood, Asbestos, Mica, Ceramics, Glass - properties and applications of PVC- effects of the (i) Filler (ii) Stabilizer (iii) Plasticizer (iv) Additives on P.V.C. - Permittivity of different Di - electric materials- Polarization - Dielectric Loss – Applications of Dielectrics.

### **3. Magnetic Materials**

Classification of magnetic materials - Soft & Hard magnetic materials- B-H Curves - Hysteresis loop - Hysteresis loss - Steinmetz equation - Eddy Current Loss -- Curie Point – Magnetostriction.

### **4. Magnetic effects of Electric Current and Electromagnetic Induction**

Coulombs laws- Permeability - Lines of force – Right Hand Thumb rule - Field pattern due to (i) long straight current carrying conductor (ii) solenoid -Work Law and its applications – Biot Savart's Law (Laplace Law)- Mechanical force on a current carrying conductor placed inside a magnetic field - Direction of force - Fleming's Left Hand rule- Magnetic circuit- mmf- Flux - Reluctance - Comparison of Magnetic circuit With Electric circuit - Magnetic leakage flux and leakage Co-efficient. Faraday's laws – Dynamically and Statically induced E.M.F-Lenz's Law & Fleming's Right Hand rule -Self and Mutual inductance - Co-efficient of coupling - Energy stored in a magnetic field.

### **5. Electrostatics and Capacitance**

Coulomb's Laws of Electrostatics–Permittivity, Electric flux, Flux density, Field intensity - Electrostatic field patterns due to (i) Isolated positive charge (ii) Isolated negative charge(iii)Unlike

charges placed side by side (iv) Like charges placed side by side -Comparison of Electrostatic and Magnetic lines of force - Concept of Electric potential and Potential difference –Di-electric strength – Di-electric constant - Capacitance – Factors affecting the Capacitance of Capacitor– capacitance of a parallel plate capacitor - Types of Capacitors- Equivalent capacitance when connected in series and parallel - Uses- Energy stored in a Capacitor.

#### REFERENCE BOOKS

- 1 Dr.K.Padmanabham - Electronic Components – Laxmi publications (P) Ltd.
- 2 Electrical Engineering Materials – N.I T.T.T.R Publications
- 3 B.K.Agarwal - Introduction to Engineering materials – Tata McGraw Hill Publishers
- 4 Ian P.Jones- Material science for Electrical and Electronic Engineers – Oxford Publications
- 5 B.L.Theraja – Electrical Technology, Vol.-1 – S.Chand& Co. Publications

#### Syllabus to be Covered for Unit Tests

Unit Test	Learning Outcomes to be Covered
Unit Test – I	From 1.1 to 2.6
Unit Test – II	From 2.7 to 4.6
Unit Test – III	From 4.7 to 5.12

### BASIC ELECTRICAL TECHNOLOGY

Course code	Course title	No. of periods/week	Total no. of Periods	Marks for FA	Marks for SA
EE-106	BASIC ELECTRICAL TECHNOLOGY	06	180	20	80

Chapter No.	Title	No. of Periods	Weightage	No. of short question (3 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Basic Principles of Electricity	35	19	3	1	CO1
2	Resistive Circuits	40	23	1	2	CO2
3	Work, Power, Energy and Heating effects of Electric Current	40	29	3	2	CO3
4	Conversion Techniques	25	13	1	1	CO4
5	Network Theorems	40	26	2	2	CO5
<b>TOTAL</b>		<b>180</b>	<b>110</b>	<b>10</b>	<b>8</b>	

<b>COURSE OBJECTIVES</b>	i. To understand the basic principles of Electricity and analysing resistive circuits. ii. To comprehend the heating effects of electric current iii. To analyse various DC network theorems.
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<b>COURSE OUTCOMES</b>	<b>CO1</b>	EE-106.1	Understand basic principles of electricity
	<b>CO2</b>	EE-106.2	Familiarise with various laws and analysis of resistive circuits
	<b>CO3</b>	EE-106.3	Understand work, power, energy concepts and heating effects of Electric current
	<b>CO4</b>	EE-106.4	Analyse various electric circuit source conversion techniques
	<b>CO5</b>	EE-106.5	Analyse various DC network theorems

## LEARNING OUTCOMES

### 1. Basic Principles of Electricity

- 1.1 Define Electric Current, Potential difference, Voltage and EMF
- 1.2 State Ohm's Law and solve problems
- 1.3 List the limitations of Ohm's Law
- 1.4 Define the terms
  - i) Specific resistance
  - ii) Conductance
  - iii) Conductivity
- 1.5 Derive the relation  $R = \rho l/a$  and solve the problems
- 1.6 Explain the effects of temperature on resistance
- 1.7 Develop the expression for resistance at any temperature as  $R_t = R_o (1 + \alpha_o t)$
- 1.8 Define temperature Co-efficient of resistance and give its unit
- 1.9 Write the formula for Co-efficient of resistance at any temperatures  $\alpha_t = \frac{\alpha_o}{1 + \alpha_o t}$

### 2. Resistive Circuits

- 1.1 Develop the expressions for equivalent Resistance with simple SERIES and PARALLEL connections
- 1.2 Solve problems on equivalent resistance in case of Series- Parallel networks
- 1.3 State the concept of division of current when two Resistors are connected in parallel and solve the problems
- 1.4 Differentiate between active and passive circuits.
- 1.5 Define junction, branch and loop in circuits
- 1.6 State (i) Kirchhoff's current law (KCL) (ii) Kirchhoff's voltage law (KVL)
- 1.7 Solve problems by applying branch current method only

### 3. Work, Power, Energy and Heating effects of Electric Current

- 3.1 State and explain electric Work, Power and Energy
- 3.2 Solve problems on Work, Power and Energy
- 3.3 Mention the typical power ratings of home appliances like Electric lamps (Incandescent, fluorescent, CFL & LED), Water heater, Electric Iron, Fans, Refrigerators, Air and Water coolers, Television sets, Air Conditioners, Water Pumps, Computers, Printers etc.
- 3.4 Calculate Electricity bill of domestic consumers as per the Electricity Tariff
- 3.5 Define Joule's law and state its expression.
- 3.6 Define Thermal efficiency
- 3.7 Solve problems on Electric heating
- 3.8 Explain the applications of heat produced due to Electric current in
  - (i) Metal Filament lamp
  - (ii) Electric kettle
  - (iii) Electric Cooker
  - (iv) Geyser

### 4. Conversion Techniques

- 4.1 Explain star and delta circuits
- 4.2 Explain the concept of circuit transformation and equivalent circuits
- 4.3 Develop transformation formulae for star- delta transformations and vice-versa
- 4.4 Solve problems on Star Delta Transformation.
- 4.5 Explain ideal voltage source & ideal current source
- 4.6 Explain Source transformation technique
- 4.7 Solve simple problems on Source transformation technique

## 5. Network Theorems

- 5.1 State the need for network theorems
- 5.2 list different types of theorems applied to DC circuits
- 5.3 State and explain Super position theorem.
- 5.4 State and explain Thevenin's theorem.
- 5.5 State and explain Norton's theorem
- 5.6 State and explain Maximum power transfer theorem.
- 5.7 Solve problems on the above theorems (All the theorems with reference to D.C only)

## HYPONATED COURSE CONTENTS

### 1. Basic Principles of Electricity

Electric current – Electric Potential, Potential difference, voltage and EMF - Ohm's law and its limitations –Resistance– Specific Resistance –Conductance- Conductivity – effects of temperature on resistance- Temperature coefficient of Resistance.

### 2. Resistive Circuits

Resistances in series, parallel and series-parallel combinations -concept of division of current when two Resistors are connected in parallel - active and passive circuits-junction, branch, loop – KCL& KVL.

### 3. Work, Power & Energy and Magnetic Effects of Electric Current

Units of electric Work, Power and Energy – Ratings of different Domestic Appliances -Calculation of Electricity bill of Domestic Consumer – Mechanical Equivalent of Heat (Joules Law) -Thermal Efficiency - Heat produced due to flow of Current in Metal Filament lamps, Electrical Kettle, Electric Cooker, Geyser

### 4. Conversion Techniques

Star and delta circuits-ideal voltage source & ideal current source -source transformation- Star-Delta& Delta-Star Transformations.

### 5. Network Theorems

Need for network theorems - superposition theorem - Thevenin's Theorem - Norton's theorem - Maximum transfer theorem.

## REFERENCE BOOKS

- |   |   |   |
|---|---|---|
| 1) B.L.Theraja – Electrical Technology, Vol.-1    | – | S.Chand & Co. Publications                          |
| 2) V. K .Mehta                                    | - | Introduction to Electrical Engg                     |
| 3) J.B.Gupta                                      | – | A course in Electrical Technology –<br>KATSON BOOKS |
| 4) G.B.Bharadhwajan& A. SubbaRao                  | - | Elements of Electrical Engineering.                 |
| 5) William H. Hayt – Engineering Circuit Analysis | – | Tata McGraw - Hill                                  |

### Syllabus to be Covered for Unit Tests

Unit Test	Learning Outcomes to be Covered
Unit Test – I	From 1.1 to 2.4
Unit Test – II	From 2.5 to 4.2
Unit Test – III	From 4.3 to 5.7

## ENGINEERING DRAWING

Course code	Course Title	No. of periods /week	Total No. of periods	Marks for FA	Marks for SA
EE-107	ENGINEERING DRAWING	3	90	40	60

Chapter No.	Title	No. of Periods	Weightage	No. of short question (5 marks)	No. of Essay questions (10 marks)	CO'S Mapped
1	Use of Drawing Instruments, Free Hand Lettering and Dimensioning Practice	10	10	2	0	CO1
2	Principles of Geometric Constructions	15	15	1	1	CO2
3	Projections of points, lines, planes and solids	20	25	1	2	CO3
4	Sectional Views	20	10	0	1	CO4
5	Orthographic projection	25	20	0	2	CO5
<b>TOTAL</b>		<b>90</b>	<b>80</b>	<b>4</b>	<b>6</b>	

<b>COURSE OBJECTIVES</b>	Upon completion of the course the student shall be able to understand the basic graphic skills and use them in preparation, reading and interpretation of engineering drawings.
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COURSE OUTCOMES	CO1	EE-107.1	Practice the use of engineering drawing instruments and Familiarise with the conventions to be followed in engineering drawing as per BIS
	CO2	EE-107.2	Construct the i) basic geometrical constructions ii) engineering curves
	CO3	EE-107.3	Visualise and draw the projections of i) Points ii) Lines iii) Regular Planes iv) Regular Solids
	CO4	EE-107.4	Visualise and draw the sectional views of components
	CO5	EE-107.5	Visualise and draw the orthographic projections of components

## LEARNING OUTCOMES

Upon completion of the course the student shall able to

### 1.0 Use of Drawing Instruments, Free Hand Lettering and Dimensioning Practice

- 1.1 State the importance of drawing as an engineering communication medium
- 1.2 Select the correct instruments to draw the different lines / curves.
- 1.3 Use correct grade of pencil and other instruments to draw different types of lines and for different purposes
- 1.4 Identify the steps to be taken to keep the drawing clean and tidy.
- 1.5 Write titles using vertical and slopping (inclined) lettering and numerals of 7mm, 10mm and 14mm height.
- 1.6 Acquaint with the conventions, notations, rules and methods of dimensioning in engineering drawing as per the B.I.S.
- 1.7 Dimension a given drawing using standard notations and desired system of dimensioning.

### 2.0 Principles of Geometric Constructions

- 2.1 Practice the basic geometric constructions like i) dividing a line into equal parts  
i) Exterior and interior tangents to the given two circles  
ii) Tangent arcs to two given lines and arcs
- 2.2 Draw any regular polygon using general method when i) side length is given  
i) Inscribing circle radius is given ii) describing circle radius is given
- 2.3 Draw the engineering curves like i) involute ii) cycloid

### 3.0 Projections of points, lines, planes and solids (All in first quadrant only)

- 3.1 Explain the basic principles of the orthographic projections
- 3.2 Visualise and draw the projection of a point with respect to reference planes (HP&VP)
- 3.3 Visualise and draw the projections of straight lines with respect to two reference Planes (up to lines parallel to one plane and inclined to other plane)
- 3.4 Visualise and draw the projections of planes (up to planes perpendicular to one plane and inclined to other plane)
- 3.5 Visualise and draw the projections of regular solids like Prisms, Pyramids, Cylinder, Cone(up to axis of solids parallel to one plane and inclined to other plane)

### 4.0 Sectional Views

- 4.1 Identify the need to draw sectional views.
- 4.2 Draw sectional views of regular solids by applying the principles of hatching.

### 5.0 Orthographic projection

- 5.1 Draw the orthographic views of an object from its pictorial drawing.
- 5.2 Draw the minimum number of views needed to represent a given object fully.

### Competencies and Key competencies to be achieved by the student

S.No	Major topic	Key Competency
1.	Use of Drawing Instruments, Free Hand Lettering and Dimensioning Practice	<ul style="list-style-type: none"> <li>Explain the linkages between Engineering drawing and other subjects of study in Diploma course.</li> </ul>
		<ul style="list-style-type: none"> <li>Select the correct instruments to draw various entities in different orientation</li> </ul>
		<ul style="list-style-type: none"> <li>Write titles using sloping and vertical lettering and numerals as per B.I.S (Bureau of Indian standards)</li> </ul>
		<ul style="list-style-type: none"> <li>Dimension a given drawing using standard notations and desired system of dimensioning</li> </ul>
2.	Geometrical construction	<ul style="list-style-type: none"> <li>Dividing a line into equal parts, tangents to circles, Construct involute, cycloid from the given data.</li> </ul>
3.	Projection of points, Lines, Planes & Solids	<ul style="list-style-type: none"> <li>Draw the projections of points, straight lines, planes &amp; solids with respect to reference planes (HP&amp; VP)</li> </ul>
4.	Sectional Views	<ul style="list-style-type: none"> <li>Differentiate between true shape and apparent shape of section</li> <li>Apply principles of hatching.</li> <li>Draw simple sections of regular solids</li> </ul>
5.	Orthographic Projection	<ul style="list-style-type: none"> <li>Draw the minimum number of views needed to represent a given object fully.</li> </ul>

#### COURSE CONTENTS:

- NOTES:**
- B.I.S Specification should invariably be followed in all the topics.
  - A-3 Size Drawing Sheets are to be used for all Drawing Practice Exercises.

#### 1.0 Use of Drawing Instruments, Free Hand Lettering and Dimensioning Practice

Explanation of the scope and objectives of the subject of Engineering Drawing . Its importance as a graphic communication -Need for preparing drawing as per standards – SP-46 –1988 – Mention B.I.S - Role of drawing in -engineering education - Basic Tools, tools for drawing– Mentioning of names under each classification and their brief description -Scales: Recommended scales reduced & enlarged -Lines: Types of lines, selection of line thickness - Selection of Pencils -Sheet Sizes: A0, A1, A2, A3, A4, A5, Layout of drawing sheets in respect of A0, A1, A3 sizes, Sizes of the Title block and its contents - Care and maintenance of Drawing Sheet,

Importance of lettering – Types of lettering -Guide Lines for Lettering Practicing of letters & numbers of given sizes (7mm, 10mm and 14mm)-Advantages of single stroke or simple style of lettering - Use of lettering stencils- Purpose of engineering Drawing, Need of B.I.S code in dimensioning -Shape description of an Engineering object -Definition of Dimensioning size description -Location of features, surface finish, fully dimensioned Drawing -Notations or tools of dimensioning, dimension line extension line, leader line, arrows, symbols, number and notes,

rules to be observed in the use of above tools -Placing dimensions: Aligned system and unidirectional system ( SP-46-1988)-Arrangement of dimensions Chain, parallel, combined progressive, and dimensioning by co-ordinate methods-The rules for dimensioning standard, features “Circles (holes) arcs, angles, tapers, chamfers, and dimension of narrow spaces.

## **2.0 Geometric Constructions**

Division of a straight line into given number of equal parts –Drawing interior and exterior tangents to two circles of given radii and centre distance-Drawing tangent arc of given radius to touch two lines inclined at given angle (acute, right and obtuse angles), Tangent arc of given radius touching a circle or an arc and a given line, Tangent arcs of radius R, touching two given circles internally and externally-Construction of any regular polygon by general method for given side length, inscribing circle radius and describing/superscripting circle radius - Involute, Cycloid, explanations as locus of a moving point, their engineering application, viz., Gear tooth profile, screw threads, springs etc. – their construction

## **3.0 Projection of points, lines and planes and Solids (All in first quadrant only)**

Classification of projections, Observer, Object, Projectors, Projection, Reference Planes, Reference Line, Various angles of projections –Differences between first angle and third angle projections

Projections of points -Projections of straight line –(a) Parallel to both the planes, (b)Perpendicular to one of the planes and (c) Inclined to one plane and parallel to other planes-Projections of regular planes-(a) Plane parallel to one of the reference planes, (b) Plane perpendicular to HP and inclined to VP and vice versa- Projections of regular solids- (a) Axis perpendicular to one of the planes, (b) Axis parallel to VP and inclined to HP and vice versa.

## **4.0 Sectional Views**

Need for drawing sectional views – what is a sectional view - Hatching – Section of regular solids inclined to one plane and parallel to other plane.

## **5.0 Orthographic Projections**

Meaning of orthographic projection - Using a viewing box and a model – Number of views obtained on the six faces of the box, - Legible sketches of only 3 views for describing object - Concept of front view, top view, and side view sketching these views for a number of engineering objects - Explanation of first angle projection. – Positioning of three views in First angle projection -Projection of points as a means of locating the corners of the surfaces of an object – Use of meter line in drawing a third view when other two views are given -Method of representing hidden lines -Selection of minimum number of views to describe an object fully.

## **REFERENCE BOOKS**

- 1 Engineering Graphics by P I Varghese – ( McGraw-hill)
- 2 Engineering Drawing by Basant Agarwal & C.M Agarwal - ( McGraw-hill)
- 3 Engineering Drawing by N.D.Bhatt.
- 4 T.S.M. & S.S.M on “ Technical Drawing” prepared by T.T.T.I., Madras.
- 5 SP-46-1998 – Bureau of Indian Standards.

**Table specifying syllabus to be covered for UNIT TEST I, II and III.**

<b>Unit Test</b>	<b>Learning Outcomes to be Covered</b>
Unit Test – I	From 1.1 to 2.3
Unit Test – II	From 3.1 to 3.5
Unit Test – III	From 4.1 to 5.2

## ELECTRICAL WIRING LABORATORY

Course code	Course title	No. of periods/ week	Total no.of periods	Marks for FA	Marks for SA
EE-108	ELECTRICAL WIRING LABORATORY	06	180	40	60

S.no	Chapter Title	No.of Periods	CO'S Mapped
1	Wiring tools and Accessories	20	CO1
2	Electrical Wiring Joints and Soldering Practice	30	CO2
3	Lamp Circuits	50	CO3
4	DC and AC circuits	40	CO4
5	Test and repair of domestic appliances	40	CO5
<b>TOTAL</b>		<b>180</b>	

<b>COURSE OBJECTIVES</b>	1) To familiarise with the knowledge of different wiring tools used in electrical wiring 2) To know the etiquette of working in the domestic wiring 3) To identify and rectify the simple faults that can occur in domestic appliances
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<b>COURSE OUTCOMES</b>	CO1	EE-108.1	Understanding various tools and know their usage
	CO2	EE-108.2	Perform different joints, Soldering practice and execute different wiring circuits
	CO3	EE-108.3	Perform various lamp control methods
	CO4	EE-108.4	Identify the difference between DC and AC
	CO5	EE-108.5	Testing and repairing of domestic applications.

## LEARNING OUTCOMES

### 1. Wiring Tools and Accessories

- 1.1 Identify the following electrical wiring tools with respect to
  - i) Size ii) Shape iii) Purpose iv) Speed v) Use
  - a) Screw drivers
  - b) Pliers
  - c) Drilling machines & Drilling Bits.
  - d) Rawl plug jumper, and poker
  - e) Voltage/line tester
  - f) Splicers (insulation remover)
  - g) Standard Wire gauge
- 1.2 Identify different types of Electrical Wiring accessories with respect to
  - i) Size ii) Shape iii) Purpose iv) Use.
  - a) Switches
  - b) Ceiling roses
  - c) Lamp Holders and Adopters
  - d) Sockets
  - e) Plug
  - f) Fuses
- 1.3 Identify different types of main switches with respect to
  - i) Rating ii) Purpose iii) Use.SP, DP mains, TP, ICDP, ICTP, SPDT, DPDT, TPDT, Changeover-Knife type, Rotary, Micro, Modular switches, 2-pole and 3-pole MCBs
- 1.4 Study different types of wires and cables (1/18, 3/20, 7/20) with respect to sizes rating, purpose and use etc

### 2. Electrical Wiring Joints and Soldering Practice

- 2.1 Prepare Straight joint/ Married joint
- 2.2 Prepare T joint
- 2.3 Prepare Western union joint
- 2.4 Prepare Pigtail joint
- 2.5 Familiarisation to use soldering tools and components
- 2.6 Soldering simple electronic circuits on PCB

### 3. Lamp Circuits

- 3.1 Make a circuit with One lamp controlled by one switch using PVC surface conduit system
- 3.2 Make a circuit with Two lamps controlled by two switches using PVC surface conduit system
- 3.3 Make a circuit with One lamp controlled by one switch and provision of 2/3-pin socket.
- 3.4 Make a circuit for Stair-case wiring
- 3.5 Make a circuit for Go-down wiring
- 3.6 Control two Lamps by Series - Parallel connection using one 1-way switch & two 2-way switches with PVC surface conduit system
- 3.7 Control two sub-circuits through Energy meter, MCB's and two 1-way switches.
- 3.8 Prepare switch board with star delta starter, MCB, Pilot lamps for 3 phase motor
- 3.9 Control and practice the wiring for Fluorescent Lamp
- 3.10 Connect Computer by main switch board with a miniature circuit breaker.

### 4. DC and AC circuits

- 4.1 Demonstrate unidirectional current flow with 12 V battery
- 4.2 Determine polarity using a Voltmeter/LED
- 4.3 Demonstrate AC using a Low voltage Transformer
- 4.4 Practice Series and Parallel connection of Lamps
- 4.5 Practice Bright and Dim light arrangement

## 5. Test and repair of the Domestic appliances

- 5.1 Testing and repair of electric heater
- 5.2 Testing and repair of iron box
- 5.3 Testing and repair of electric kettle
- 5.4 Testing and repair of electric cooker
- 5.5 Testing and repair of electric geyser

### Competencies to be achieved by the Student:

S.No	Competencies	Key Competencies
1	Handle the different wiring tools and accessories a) Select switches, and MCB's b) Identify wires and cables as per the requirements of the load.	<ul style="list-style-type: none"> <li>• Identify the size and specifications of various tools used for electrical wiring.</li> <li>• Understand the usage of the standard wire gauge.</li> <li>• Identify the type, size and specifications of DP mains,</li> </ul>
2.1	To prepare a Straight joint/Married joint using a 7/20 Al. Cable	<ul style="list-style-type: none"> <li>• Identify the size of the cable</li> <li>• Perform splicing of Insulation properly.</li> <li>• Perform Straight joint/Married joint</li> </ul>
2.2	To prepare a T joint using a 7/20 Al. Cable	<ul style="list-style-type: none"> <li>• Insert the leads of the wires properly as per the sketches.</li> <li>• Twist the wires properly.</li> </ul>
2.3	To prepare a Western union joint using a single strand Al. Cable	<ul style="list-style-type: none"> <li>• Overlap the two wires properly</li> <li>• Twist the binding wires properly</li> </ul>
2.4	To prepare a Pig tail joint using a single strand Copper Cable	<ul style="list-style-type: none"> <li>• Place the wires in V-shape.</li> <li>• Twist the wires in clock wise direction.</li> </ul>
2.5	To Familiarise various soldering tools and components	<ul style="list-style-type: none"> <li>• Identifying Soldering gun, flux, lead</li> </ul>
2.6	To solder simple electronic circuits on PCB	<ul style="list-style-type: none"> <li>• Draw the layout of circuit</li> <li>• Carefully Soldering the circuit on PCB.</li> </ul>
3.1	To control one lamp by one 1-way switch with PVC surface conduit wiring system	<ul style="list-style-type: none"> <li>• Draw wiring diagram</li> <li>• Identify the size of cable, PVC pipe, type of 1-way switch and lamp holder.</li> <li>• Make Connections as per Wiring Diagram</li> </ul>

3.2	To control two lamps by two 1-way switches with PVC surface conduit wiring system	<ul style="list-style-type: none"> <li>• Draw wiring diagram</li> <li>• Handle the screw driver, electrician Knife, line tester to fix the PVC pipe using saddles and junction boxes.</li> <li>• Select colour and length of wire for phase and neutral</li> <li>• Switch on the supply after making of the connections</li> <li>• Disconnect the circuit after testing.</li> </ul>
3.3	To control one lamp and 2/3 pin socket by two 1-way switches with PVC surface conduit wiring system	<ul style="list-style-type: none"> <li>• Connect 2/3 pin socket properly with respect to phase, neutral and earth.</li> <li>• Connect phase wire through switches.</li> </ul>
3.4	Stair-case wiring	<ul style="list-style-type: none"> <li>• Select two 2-way switches</li> <li>• Connect 2-way switches as per circuit diagram.</li> <li>• Test with 1-phase, 230V, 50 Hz supply to the circuit connected through ICDP switch.</li> </ul>
3.5	Go-down wiring scheme	<ul style="list-style-type: none"> <li>• Draw wiring diagram</li> <li>• Connect the circuit as per the diagram.</li> <li>• Observe sequence of operation of switches</li> <li>• Test with 1-phase, 230 V, 50 Hz supply to the circuit, neutral wire to the bottom point of the 1-way switch and phase to the first point of lamp holder</li> </ul>
3.6	Series-Parallel connection	<ul style="list-style-type: none"> <li>• Select colour and length of wire for phase and neutral.</li> <li>• Make connections as per wiring diagram.</li> <li>• Draw wire through PVC pipe properly</li> <li>• Observe glow intensity of lamps for series and parallel connections</li> </ul>
3.7	Control two sub circuits through Energy meter, MCB's and two 1-way switches	<ul style="list-style-type: none"> <li>• Draw wiring diagram.</li> <li>• Identify the size of cable, 1-way switch, PVC pipe, MCB, capacity of Inverter and Socket</li> <li>• Read the specifications of MCB, capacity of Inverter and Socket</li> <li>• Make connections as per wiring diagram.</li> <li>• Draw wire through PVC pipe properly.</li> <li>• Connect supply to Inverter through MCB properly.</li> <li>• Select appropriate socket with switch control.</li> <li>• Make earth wire connections for required points.</li> </ul>

3.8	Prepare switch board with star delta starter, MCB, Pilot lamps for 3 phase motor	<ul style="list-style-type: none"> <li>• Select the size of cable, PVC pipe, star-delta starter, MCB and lamp holder</li> <li>• Make connections as per wiring diagram.</li> <li>• Draw wire through PVC pipe properly.</li> <li>• Draw wire of the 3-phase to the motor through star-delta starter.</li> <li>• Test with 3-phase, 415 V, 50 Hz supply to the circuit connected through ICDP switch.</li> <li>• Test by changing any two phases of input supply</li> </ul>
3.9	Wiring practice of fluorescent lamp	<ul style="list-style-type: none"> <li>• Make connections as per wiring diagram.</li> <li>• Connect top point and bottom point of the choke to tube light properly.</li> <li>• Note the importance and working of starter.</li> </ul>
3.10	Connect computer by main switch board with a miniature circuit breaker.	<ul style="list-style-type: none"> <li>• Draw wiring diagram.</li> <li>• Identify the size of cable, 1-way switch, PVC pipe, MCB and Sockets</li> <li>• Read the specifications of MCB and Sockets</li> <li>• Make connections as per wiring diagram.</li> <li>• Connect supply to Computer through MCB properly.</li> <li>• Select appropriate sockets with 1-way switch control.</li> <li>• Make earth wire connections for require points.</li> </ul>
4.1	Demonstrate difference between DC and AC	<ul style="list-style-type: none"> <li>• Connect DC source and measure V &amp; I</li> <li>• Connect proper AC source and measure V &amp; I</li> <li>• Make inferences.</li> </ul>
5.1	Testing and repair of domestic appliances	<ul style="list-style-type: none"> <li>• Inspect the appliance visually.</li> <li>• Check for any discrepancies.</li> <li>• Perform the disassembling operation</li> <li>• Test the inner parts for any faults</li> <li>• Rectify the faults if any.</li> <li>• Replace the parts if necessary.</li> <li>• Perform the assembling.</li> <li>• Test the Domestic appliance for proper functioning.</li> </ul>

**Note:**

1. Every student has to bring insulated tool kit and follow the general safety precautions throughout the lab sessions
2. Should not touch the live terminals.

**PHYSICS LAB**

<b>Course code</b>	<b>Course title</b>	<b>No. of periods/ week</b>	<b>Total no. of periods</b>	<b>Marks for FA</b>	<b>Marks for SA</b>
EE-109	PHYSICS LAB	1.5	45	20	30

<b>S.No</b>	<b>List of experiments</b>	<b>No.of Periods</b>	<b>CO's Mapped</b>
1.	Vernier calipers	03	CO1
2.	Micrometer (Screw gauge)	03	
3.	Verification of Lami's theorem using concurrent forces	03	
4.	Determination of 'g' using simple pendulum	03	CO2
5.	Focal length and focal power of convex lens	03	
6.	Refractive index of solid using travelling microscope	03	
7.	Verification of Boyle's law using Quill tube	03	CO3
8	Determination of pole strength of the bar magnet through magnetic field lines	03	
9	Resonance apparatus – Determination of velocity of sound in air	03	
	<b>Experiments for demonstration</b>		
10	Meter bridge – Determination of resistance and specific resistance of a wire	03	CO4
11	Verification of Newton's law of cooling	03	
12	Photo electric cell – Study of its characteristics	03	
	Revision	06	
	Test	03	
	<b>Total:</b>	<b>45</b>	

<b>COURSE OBJECTIVES</b>	<p>(1) To provide strong practical knowledge of Physics to serve as a tool for various device applications in Engineering.</p> <p>(2) To enhance scientific skills of the students by incorporating new experiments so as to enrich the technical expertise of the students as required for industries.</p>
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<b>COURSE OUTCOMES</b>	CO1	Improving accuracy in various measurements; understanding the nature of the forces keeping the body in equilibrium.
	CO2	Estimating the acceleration caused by the gravity of earth; Practical study of the concepts of refraction of light at curved/plane surface.
	CO3	Understanding the pressure of the gas as function of its volume; study of the combined magnetic field of the earth and an artificial magnet to estimate its pole strength; Estimating the velocity of sound in air through resonance phenomenon.
	CO4	Applying Kirchoff's laws to evaluate the specific resistance of a wire; Study of exchange of heat from system to surrounding by graphical analysis, Conversion of light energy to micro currents as potential engineering application.

#### CO-PO MAPPING

CO	PO1	PO2	PO3	PO4	PO5	PO6	PO7
CO1	3	2	2	2	2	1	2
CO2	3		1	1	1	1	1
CO3	3	2			1		
CO4	3	2	2			1	2

### CO-PO Mapping Strength

EE -109	PHYSICS LAB No of Course Objectives : 4				No of periods 45
POs	Mapped with CO No	Methods addressing PO in Col 1 NO	%	1,2,3	remarks
PO1	CO1,CO2,CO3,CO4	15	33.3 %	2	>40% level 3 (highly addressed)  25% to 40% level 2 (moderately addressed)  5% to 25% level 1 (Low addressed)  < 5% (not addressed)
PO2	CO1,CO3, CO4	8	17.8%	1	
PO3	CO1, CO2, CO4	6	13.3%	1	
PO4	CO1, CO2	3	6.7%	1	
PO5	CO1,CO2, CO3	5	11.1%	1	
PO6	CO1, CO2, CO4	3	6.7%	1	
PO7	CO1, CO2, CO4	5	11.1%	1	

3 = strongly mapped, 2 = moderately mapped, 1 = slightly mapped

**Note: The gaps in CO and PO mapping will be achieved by one or more appropriate activities from the following.**

- |                          |                       |                               |
|--------------------------|-----------------------|-------------------------------|
| (i) Seminars             | (ii) Viva-voce        | (iii) Assignments             |
| (iv) Quiz competitions   | (v) Industrial visits | (vi) Tech fest                |
| (viii) Group discussions | (ix) Virtual labs     | (vii) Mini project            |
|                          |                       | (x) Library visit for e-books |

#### LEARNING OUTCOMES

**Upon completion of the course the student shall be able to**

- 1.0 Practice with Vernier calipers to determine the volumes of cylinder and sphere.
- 2.0 Practice with Screw gauge to determine thickness of a glass plate and cross sectional area of a wire.
- 3.0 Verify the Lami's theorem using concurrent forces.
- 4.0 Determine the value of acceleration due to gravity (g) using Simple Pendulum. To verify the result from  $l-T^2$  graph.
- 5.0 Calculate the Focal length and focal power of convex lens using distant object method and U-V method. To verify the result from U-V graph and  $1/U - 1/V$  graph methods.
- 6.0 Determine the refractive index of a solid using travelling microscope
- 7.0 Verify the Boyle's law using Quill tube. To draw a graph between P and  $1/l$ .

- 8.0 Determination of magnetic pole strength of a bar magnet by drawing magnetic lines of force and locating null points (either N - N or N - S method)
- 9.0 Determine the velocity of sound in air at room temperature and its value at zero degree Centigrade using resonance apparatus.
- 10.0 Determine the resistance and specific resistance of material of a wire using Meter Bridge
- 11.0 To verify the Newton's law of cooling.
- 12.0 To study the characteristics of photo electric cell.

**Competencies and Key competencies to be achieved by the student**

Name of the Experiment (Nu	Competencies	Key competencies
1 . Practice on Vernier Calipers (03)	<ul style="list-style-type: none"> <li>• Find the Least count</li> <li>• Fix the specimen in position</li> <li>• Read the scales</li> <li>• Calculate the physical quantities of given object</li> </ul>	<ul style="list-style-type: none"> <li>• Read the scales</li> <li>• Calculate the requisite physical quantities of given objects</li> <li>• Calculating volumes of the cylinder and sphere</li> </ul>
2. Practice on Screw gauge(03)	<ul style="list-style-type: none"> <li>• Find the Least count</li> <li>• Fix the specimen in position</li> <li>• Read the scales</li> <li>• Calculate thickness of glass plate and cross section of wire from radius</li> </ul>	<ul style="list-style-type: none"> <li>• Read the scales</li> <li>• Noting zero error</li> <li>• Calculate thickness of given glass plate</li> <li>• Calculate cross section of wire from radius</li> </ul>
3. Verification of Lami's theorem forces(03)	<ul style="list-style-type: none"> <li>• Making experimental set up</li> <li>• Fix suitable weights</li> <li>• Note the positions of threads on drawing sheet</li> <li>• Find the angles between the concurrent forces</li> <li>• Changing weights appropriately</li> <li>• Verify Lami's theorem</li> </ul>	<ul style="list-style-type: none"> <li>• Measuring angles between the forces</li> <li>• Marking the directions forces on a paper               <ul style="list-style-type: none"> <li>• Verifying Lami's theorem from the weights and measured angles between the forces.</li> </ul> </li> </ul>

4. Simple pendulum(03)	<ul style="list-style-type: none"> <li>• Fix the simple pendulum to the stand</li> <li>• Adjust the length of pendulum</li> <li>• Find the time for number of oscillations (say 20)</li> <li>• Find the time period</li> <li>• Calculate the acceleration due to gravity</li> <li>• Draw <math>l-T^2</math> graph</li> </ul>	<ul style="list-style-type: none"> <li>• Find the time for number of oscillations</li> <li>• Find the time period</li> <li>• Calculate the acceleration due to gravity</li> <li>• Verify form <math>l-T^2</math> graph</li> </ul>
5. Focal length and Focal power of convex lens (03)	<ul style="list-style-type: none"> <li>• Fix the object distance</li> <li>• Find the Image distance</li> <li>• Calculate the focal length and power of convex lens</li> <li>• Draw <math>u-v</math> and <math>1/u - 1/v</math> graphs</li> </ul>	<ul style="list-style-type: none"> <li>• Find focal length from distant object method.</li> <li>• Calculate the focal length and power of convex lens</li> <li>• Verify result from <math>u-v</math> and <math>1/u - 1/v</math> graphs</li> </ul>
6 Refractive index of solid using traveling microscope(03)	<ul style="list-style-type: none"> <li>• Find the least count of Vernier on microscope</li> <li>• Place the graph paper below microscope</li> <li>• Read the scales</li> </ul>	<ul style="list-style-type: none"> <li>• Reading the scales on Microscope.</li> <li>• Finding real and apparent thickness of the slab</li> <li>• Calculate the refractive</li> </ul>
7 . Boyle's law verification (03)	<ul style="list-style-type: none"> <li>• Note the atmospheric pressure</li> <li>• Fix the Quill tube to retort stand</li> <li>• Find the length of air column</li> <li>• Find the pressure of enclosed air</li> <li>• Find and compare the calculated values of <math>P \times l</math></li> </ul>	<ul style="list-style-type: none"> <li>• Fixing Quill tube in various positions on retort stand.</li> <li>• Find the length of air column</li> <li>• Find the pressure of enclosed air</li> <li>• Find the values of <math>P \times l</math></li> <li>• Verify Boyle's law.</li> </ul>
8. Mapping of magnet lines of force (03)	<ul style="list-style-type: none"> <li>• Draw magnetic meridian</li> <li>• Place the bar magnet in N-N or N-S directions</li> <li>• Draw magnetic lines of force</li> <li>• Locate the neutral points</li> </ul>	<ul style="list-style-type: none"> <li>• Draw the pattern of magnetic lines of force</li> <li>• Locate the neutral points</li> <li>• Calculating pole strength of the bar magnet</li> </ul>

<p>9. Velocity of sound in air – Resonance method (03)</p>	<ul style="list-style-type: none"> <li>• Arrange the resonance apparatus</li> <li>• Adjust the reservoir level for booming sound</li> <li>• Find the first and second resonating lengths</li> <li>• Calculate velocity of sound .</li> </ul>	<ul style="list-style-type: none"> <li>• Adjust the reservoir level</li> <li>• Find the first and second resonating lengths</li> <li>• Calculate velocity of sound at room temperature and at 0° C</li> </ul>
<p>10. Meter bridge(03)</p>	<ul style="list-style-type: none"> <li>• Make the circuit connections</li> <li>• Find the balancing length</li> <li>• Calculate unknown resistance</li> <li>• Find the radius of wire</li> <li>• Calculate the specific resistance</li> </ul>	<ul style="list-style-type: none"> <li>• Making connections as per circuit diagram.</li> <li>• Find the balancing length</li> <li>• Calculate unknown resistance</li> <li>• Calculate the specific resistance of the given wire</li> </ul>
<p>11. Verification of Newton’s law of Cooling (03)</p>	<ul style="list-style-type: none"> <li>• Heating liquid in a beaker using a heating element</li> <li>• Inserting thermometer in liquid in calorimeter</li> <li>• Stirring liquid</li> <li>• Measuring temperatures as a function of time using thermometer</li> <li>• Plotting a cooling curve</li> </ul>	<ul style="list-style-type: none"> <li>• Measuring temperature of a liquid as function of time.</li> <li>• Plotting a cooling curve.</li> <li>• Verifying Newton’s law of cooling.</li> </ul>
<p>12. Photo electric cell – Study of its Characteristics (03)</p>	<ul style="list-style-type: none"> <li>• Experimental set up and making connections</li> <li>• Verifying intensity of light by varying distances between light source and photocell.</li> <li>• Measuring Voltage and current values.</li> </ul>	<ul style="list-style-type: none"> <li>• Making connections for experimental set up.</li> <li>• Varying distances appropriately</li> <li>• Measuring Voltage and current values.</li> <li>• Study of V- I Characteristics form graph.</li> </ul>

**Scheme of Valuation for End Practical Examination :**

<b>Activity</b>	<b>Marks</b>
For writing, Apparatus, formulae, least count (if applicable)	5
Procedure & precautions	5
Drawing Tables	3
Readings, calculations, graph (if applicable), reporting the findings	12
Viva-voce	5
<b>Total marks</b>	<b>30</b>

**CHEMISTRY LAB**

Course code	Course title	No. of periods/ week	Total no.of periods	Marks for FA	Marks for SA
EE-110	CHEMISTRY LAB	1.5	45	20	30

S.No	Name of the Experiment	No. of Periods	Mapped with COs
1.	a) Recognition of chemical substances and solutions used in the laboratory by senses. b) Familiarization of methods for Volumetric analysis.	03	CO1
2.	Preparation of Std.Na <sub>2</sub> CO <sub>3</sub> solution and making solutions of different	03	CO1
3.	Estimation of HCl solution using Std.Na <sub>2</sub> CO <sub>3</sub> solution.	03	CO2
4.	Estimation of NaOH using Std. HCl solution.	03	CO2
5.	Determination of acidity of water sample.	03	CO2
6.	Determination of alkalinity of water sample.	03	CO2
7.	Estimation of Mohr's Salt usingStd.KMnO <sub>4</sub> . Solution.	03	CO3
8.	Estimation of Ferrous ion by using Std. K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> solution.	03	CO3
9.	Determination of total hardness of water sample using Std. EDTA solution	03	CO4
10.	Estimation of Chlorides present in water sample by using Std. AgNO <sub>3</sub> solution.	03	CO4
11.	Estimation of Dissolved Oxygen(D.O) in water sample by using Std. hypo solution.	03	CO5
12.	Determination of pH using pH meter.	03	CO5
13.	Determination of conductivity of water and adjusting ionic strength	03	CO5
14.	Determination of turbidity of water.	03	CO5
15.	Estimation of total solids present in water sample.	03	CO5
<b>Total:</b>		<b>45</b>	

<b>COURSE OUTCOMES</b>	<b>CO1</b>	Operate and practice volumetric apparatus and preparation of standard solution.
	<b>CO2</b>	Evaluate and judge the neutralization point in acid base titration.
	<b>CO3</b>	Evaluate the end point of reduction and oxidation reaction.
	<b>CO4</b>	Judge the stable end point of complex formation, stable precipitation.
	<b>CO5</b>	Judge operate and demonstrate and perform precise operations with instrument for investigation of water pollution parameters.

**PO- CO mapping**

EE-110	Chemistry Lab No. of CO's:5				No. of periods : 45
Pos	Mapped with CO No.	CO periods addressing PO in Col. No. 1	%	Level 1,2,3	Remarks
PO1	CO1,CO2,CO3, CO4,CO5	12	26.66	2	>40% Level 3 (highly addressed)  25% to 40% Level 2 (moderately addressed)  5% to 25% Level1 (Low addressed)  < 5%(not addressed)
PO2	CO1,CO2,CO3, CO4,CO5	9	20	1	
PO3					
PO4	CO1,CO2,CO3, CO4,CO5	12	26.66	2	
PO5	CO2,CO3, CO4,CO5	12	26.66	2	
PO6	-	-	-	-	
PO7	-	-	-	-	

**COs-POs mapping strength (as per given table)**

CO	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PSO1	PSO2	PSO3
CO1	2	1	-	2	-	-	-	-	-	-
CO2	2	1	-	2	2	-	-	-	-	-
CO3	2	1	-	2	2	-	-	-	-	-
CO4	2	1	-	2	2	-	-	-	-	-
CO5	2	1	-	2	2	-	-	-	-	-

3=strongly mapped 2= moderately mapped 1= slightly mapped

**Note:** The gaps in CO and PO mapping will be achieved by one or more appropriate activities from the following: i) Seminars ii) Tutorials iii) Guest Lectures iv) Assignments v) Quiz competitions vi) Industrial visit vii) Tech Fest viii) Mini project ix) Group discussions x) Virtual classes xi) Library visit for e-books

## OBJECTIVES:

### Upon completion of the course the student shall be able to

- 1.0 To identify the chemical compounds and solutions by senses. Practice volumetric measurements (using pipettes, measuring jars, volumetric flask, burettes) and gravimetric measurements (using different types of balances), making dilutions, etc.
- 2.0 Practice making standard solutions with pre weighed salts and to make solutions of desired dilutions using appropriate techniques.
- 3.0 Conduct titrations adopting standard procedures and using Std.  $\text{Na}_2\text{CO}_3$  solution for estimation of HCl.
- 4.0 Conduct titrations adopting standard procedures and using Std. HCl solution for estimation of NaOH.
- 5.0 Conduct titrations adopting standard procedures to determine the acidity of given samples of water (One ground water and one surface / tap water, and rain water if available).
- 6.0 Conduct titrations adopting standard procedures to determine the alkalinity of given samples of water (One ground water and one surface / tap water).
- 7.0 Conduct titrations adopting standard procedures and using Std.  $\text{KMnO}_4$  solution for estimation of Mohr's Salt.
- 8.0 Conduct titrations adopting standard procedures and using Std.  $\text{K}_2\text{Cr}_2\text{O}_7$  solution for estimation of Ferrous ion.
- 9.0 Conduct titrations adopting standard procedures to determine the total hardness of given samples of water (One ground water and one surface / tap water) using Std. EDTA solution.
10. Conduct titrations adopting standard procedures to determine the chlorides present in the given samples of water (One ground water and one surface / tap water) using Std.  $\text{AgNO}_3$  solution.
11. Conduct the test using titrimetric / electrometric method to determine Dissolved Oxygen (D.O) in the given water samples (One sample from closed container and one from open container / tap water) by Std. Hypo solution.
12. Conduct the test on given samples of water / solutions (like soft drinks, sewage, etc.) to determine their pH using standard pH meter.
13. Conduct the test on given samples of water / solutions.
  - a) to determine conductivity.
  - b) to adjust the ionic strength of the sample to the desired value.
14. Conduct the test on given samples of solutions (coloured and non-coloured) to determine their turbidity in NTU.
15. Determine the total solids present in given samples of water (One ground water and one surface / tap water).

**Competencies and Key competencies to be achieved by the student**

Name of the Experiment (No of Periods)	Competencies	Key competencies
Recognition of chemical substances and solutions. Familiarization of methods for Volumetric analysis. (03)	-	--
Preparation of Std. Na <sub>2</sub> CO <sub>3</sub> solution and making solutions of different dilutions. (03)	<ul style="list-style-type: none"> <li>▪ Weighing the salt to the accuracy of .01 mg.</li> <li>▪ Measuring the water with volumetric flask, measuring jar, volumetric pipette and graduated pipette.</li> <li>▪ Making appropriate dilutions.</li> </ul>	<ul style="list-style-type: none"> <li>▪ Weighing the salt to the accuracy of 0.01 mg.</li> <li>▪ Measuring the water with volumetric flask, measuring jar, volumetric pipette and graduated pipette.</li> <li>▪ Making appropriate dilutions.</li> </ul>
Estimation of HCl solution using Std. Na <sub>2</sub> CO <sub>3</sub> solution. (03)	<ul style="list-style-type: none"> <li>▪ Cleaning the glassware and rinsing with appropriate solutions.</li> <li>▪ Making standard solutions.</li> <li>▪ Measuring accurately the standard solutions and titrants.</li> <li>▪ Filling the burette with titrant.</li> <li>▪ Fixing the burette to the stand.</li> <li>▪ Effectively Controlling the flow of the titrant.</li> <li>▪ Identifying the end point.</li> <li>▪ Making accurate observations.</li> <li>▪ Calculating the results.</li> </ul>	<ul style="list-style-type: none"> <li>▪ Making standard solutions.</li> <li>▪ Measuring accurately the standard solutions and titrants.</li> <li>▪ Effectively Controlling the flow of the titrant.</li> <li>▪ Identifying the end point.</li> <li>▪ Making accurate observations.</li> </ul>
Estimation of NaOH using Std. HCl solution. (03)		
Determination of acidity of water sample. (03)		
Determination of alkalinity of water sample. (03)		
Estimation of Mohr's Salt using Std. KMnO <sub>4</sub> solution. (03)		
Estimation of Ferrous ion by using Std. K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> solution (03)		
Determination of total hardness of water using Std. EDTA solution. (03)		
Estimation of Chlorides present in water sample using Std. AgNO <sub>3</sub> solution (03)		
Estimation of Dissolved Oxygen (D.O) in water sample (By titration method) (03)		
Determination of pH using pH meter. (03)	<ul style="list-style-type: none"> <li>▪ Familiarize with instrument.</li> <li>▪ Choose appropriate 'Mode' /</li> </ul>	<ul style="list-style-type: none"> <li>▪ Prepare standard solutions / buffers, etc.</li> </ul>

Determination of conductivity of water and adjusting ionic strength to required level. (03)	<p>'Unit'.</p> <ul style="list-style-type: none"> <li>▪ Prepare standard solutions / buffers, etc.</li> <li>▪ Standardize the instrument with appropriate standard solutions.</li> <li>▪ Plot the standard curve.</li> <li>▪ Make measurements accurately.</li> <li>▪ Follow Safety precautions.</li> </ul>	<ul style="list-style-type: none"> <li>▪ Standardize the instrument with appropriate standard solutions.</li> <li>▪ Plot the standard curve.</li> <li>▪ Make measurements accurately.</li> </ul>
Determination of turbidity of water. (03)		
Estimation of total solids present in water sample. (03)	<ul style="list-style-type: none"> <li>▪ Measuring the accurate volume and weight of sample.</li> <li>▪ Filtering and air drying without losing any filtrate.</li> <li>▪ Accurately weighing the filter paper, crucible and filtrate.</li> </ul>	<ul style="list-style-type: none"> <li>▪ Measuring the accurate volume and weight of sample.</li> <li>▪ Filtering and air drying without losing any filtrate.</li> <li>▪ Accurately weighing the filter paper, crucible and</li> </ul>

#### SCHEME OF VALUATION

A) Writing Chemicals, apparatus, principle and procedure.	5M
B) Demonstrated competencies.	20M
Making standard solutions.	
Measuring accurately the standard solutions and titrants.	
Effectively controlling the flow of the titrant.	
Identifying the end point.	
Making accurate observations.	
C) Viva-voce.	5M
<b>Total</b>	<b>30M</b>

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### COMPUTER FUNDAMENTALS LABORATORY

Course code	Course Title	No. of Periods/Weeks	Total No. of periods	Marks for FA	Marks for SA
EE-111	COMPUTER FUNDAMENTALS LABORATORY	3	90	40	60

#### Time schedule:

S.No.	Chapter/Unit Title	No. of sessions each of 3 periods duration	No. of Periods	CO's Mapped
1.	Computer hardware Basics	2	6	CO1
2.	Windows Operating System	2	6	CO1
3.	MS Word	8	24	CO2
4.	MS Excel	7	21	CO3
5.	MS PowerPoint	5	15	CO4
6	Adobe Photoshop	6	18	CO5
<b>Total periods</b>		<b>30</b>	<b>90</b>	

<b>COURSE OBJECTIVES</b>	i)To know Hardware Basics ii)To familiarize operating systems iii)To use MS Office effectively to enable to students use these skills in future courses iv) To use Adobe Photoshop in image editing.
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At the end of the course students will be able to			
<b>COURSE OUTCOMES</b>	CO1	CM-111.1	Identify hardware and software components
	CO2	CM-111.2	Prepare documents with given specifications using word processing software
	CO3	CM-111.3	Use Spread sheet software to make calculation and to draw various graphs / charts.
	CO4	CM-111.4	Use Power point software to develop effective presentation for a given theme or topic.
	CO5	CM-111.5	Edit digital or scanned images using Photoshop

## **LEARNING OUTCOMES:**

### **I. Computer Hardware Basics**

1. a) To Familiarize with Computer system and hardware connections  
b) To Start and Shut down Computer correctly  
c) To check the software details of the computer
2. To check the hardware present in your computer

### **II. Windows's operating system**

3. To Explore Windows Desktop
4. Working with Files and Folders
5. Windows Accessories: Calculator – Notepad – WordPad – MS Paint

### **III. Practice with MS-WORD**

6. To familiarize with Ribbon layout of MS Word  
Home – Insert - Page layout – References – Review- View.
7. To practice Word Processing Basics
8. To practice Formatting techniques
9. To insert a table of required number of rows and columns
10. To insert Objects, Clipart and Hyperlinks
11. To use Mail Merge feature of MS Word
12. To use Equations and symbols features

### **IV. Practice with MS-EXCEL**

13. To familiarize with MS-EXCEL layout
14. To access and enter data in the cells
15. To edit a spread sheet- Copy, Cut, Paste, and selecting Cells
16. To use built in functions and Formatting Data
17. To create Excel Functions, Filling Cells
18. To enter a Formula for automatic calculations
19. To sort and filter data in table.
20. To present data using Excel Graphs and Charts.
21. To develop lab reports of respective discipline.
22. To format a Worksheet in Excel, Page Setup and Print

### **V. Practice with MS-POWERPOINT**

23. To familiarize with Ribbon layout features of PowerPoint 2007.
24. To create a simple PowerPoint Presentation
25. To set up a Master Slide in PowerPoint
26. To insert Text and Objects
27. To insert a Flow Charts
28. To insert a Table
29. To insert a Charts/Graphs
30. To insert video and audio

31. To practice Animating text and objects

32. To Review presentation

#### VI. Practice with Adobe Photoshop

33. To familiarize with standard toolbox

34. To edit a photograph.

35. To insert Borders around photograph.

36. To change Background of a Photograph.

37. To change colors of Photograph.

38. To prepare a cover page for the book in your subject area.

39. To adjust the brightness and contrast of the picture so that it gives an elegant look.

40. To type a word and apply the shadow emboss effects.

#### KEY COMPETENCIES:

Expt No	Name of Experiment	Competencies	Key competencies
1 (a).	To familiarize with Computer system and hardware connections	<ul style="list-style-type: none"><li>a. Identify the parts of a Computer system: i). CPU ii). Mother Board iii) Monitor iv) CD/DVD Drive v) Power Switch vi) Start Button vii) Reset Button</li><li>b. Identify and connect various peripherals</li><li>c. Identify and connect the cables used with computer system</li><li>d. Identify various ports on CPU and connect Keyboard &amp; Mouse</li></ul>	Connect cables to external hardware and operate the computer
1 (b).	To Start and Shut down Computer correctly	<ul style="list-style-type: none"><li>a. Log in using the password</li><li>b. Start and shut down the computer</li><li>c. Use Mouse and Key Board</li></ul>	<ul style="list-style-type: none"><li>a. Login and logout as per the standard procedure</li><li>b. Operate mouse &amp; Key Board</li></ul>
1 (c).	To Explore Windows Desktop	<ul style="list-style-type: none"><li>a. Familiarize with Start Menu, Taskbar, Icons and Shortcuts</li><li>b. Access application programs using Start menu, Task manager</li><li>c. Use Help support</li></ul>	<ul style="list-style-type: none"><li>a. Access application programs using Start menu</li><li>b. Use taskbar and Task manager</li></ul>

2.	To check the software details of the computer	<ul style="list-style-type: none"> <li>a. Find the details of Operating System being used</li> <li>b. Find the details of Service Pack installed</li> </ul>	Access the properties of computer and find the details
3.	To check the hardware present in your computer	<ul style="list-style-type: none"> <li>a. Find the CPU name and clock speed</li> <li>b. Find the details of RAM and Hard disk present</li> <li>c. Access Device manager using Control Panel and check the status of devices like mouse and key board</li> <li>d. Use My Computer to check the details of Hard drives and partitions</li> <li>e. Use the Taskbar</li> </ul>	<ul style="list-style-type: none"> <li>a. Access device manager and find the details</li> <li>b. Type /Navigate the correct path and Select icon related to the details required</li> </ul>
4.	Working with Files and Folders	<ul style="list-style-type: none"> <li>a. Create folders and organizing files in different folders</li> <li>b. Use copy / paste move commands to organize files and folders</li> </ul>	<ul style="list-style-type: none"> <li>a. Create files and folders Rename, arrange and search for the required folder/file</li> </ul>
	Working with Files and Folders Continued....	<ul style="list-style-type: none"> <li>c. Arrange icons – name wise, size, type, Modified</li> <li>d. Search a file or folder and find its path</li> <li>e. Create shortcut to files and folders (in other folders) on Desktop</li> <li>f. Familiarize with the use of My Documents</li> <li>g. Familiarize with the use of Recycle Bin</li> </ul>	<ul style="list-style-type: none"> <li>b. Restore deleted files from Recycle bin</li> </ul>
5.	To use Windows Accessories: Calculator – Notepad – WordPad – MS Paint	<ul style="list-style-type: none"> <li>a. Familiarize with the use of Calculator</li> <li>b. Access Calculator using Run command</li> <li>c. Create Text Files using Notepad and WordPad and observe the difference in file size</li> </ul>	<ul style="list-style-type: none"> <li>a. Use windows accessories and select correct text editor based on the situation.</li> <li>b. Use MS paint to create /Edit pictures</li> </ul>

		d. Use MS paint and create .jpeg, .bmp files using MS Paint	and save in the required format.
6.	To familiarize with Ribbon layout of MS word. – Home – Insert- page layout- References-Review- View	a. Create/Open a document b. Use Save and Save as features c. Work on two Word documents simultaneously d. Choose correct Paper size and Printing options	a. Create a Document and name appropriately and save b. Set paper size and print options
7.	To practice Word Processing Basics	a. Typing text b. Keyboard usage c. Use mouse (Left click / Right click / Scroll) d. Use Keyboard shortcuts e. Use Find and Replace features in MS- word f. Use Undo and Redo Features g. Use spell check to correct Spellings and Grammar	a. Use key board and mouse to enter/edit text in the document. b. Use shortcuts c. Use spell check/ Grammar features for auto corrections.
8.	To practice Formatting techniques	a. Formatting Text b. Formatting Paragraphs c. Setting Tabs d. Formatting Pages e. The Styles of Word f. Insert bullets and numbers g. Themes and Templates h. Insert page numbers, header and footer	a. Format Text and paragraphs and use various text styles. b. Use bullets and numbers to create lists c. Use Templates /Themes d. Insert page numbers date, headers and footers
9.	To insert a table of required number of rows and columns	a. Edit the table by adding the fields – Deleting rows and columns –inserting sub table –marking borders. Merging and splitting of cells in a Table b. Changing the background colour of the table c. Use table design tools d. Use auto fit – fixed row/ column height/length – Even distribution of rows / columns features e. Convert Text to table and	a. Insert table in the word document and edit b. Use sort option for arranging data.

		<p>Table to Text</p> <p>f. Use Sort feature of the Table to arrange data in ascending/descending order</p>	
10.	To Insert objects, clipart and Hyperlinks	<p>a. Create a 2-page document. &amp;Insert hyperlinks and t Bookmarks.</p> <p>b. Create an organization chart</p> <p>c. Practice examples like preparing an Examination schedule notice with a hyperlink to Exam schedule table.</p>	<p>a. Insert hyperlinks &amp;Bookmarks</p> <p>b. Create organization charts/flow charts</p>
11.	To Use Mail merge feature of MS Word	<p>a. Use mail merge to prepare individually addressed letters</p> <p>b. Use mail merge to print envelopes.</p>	Use Mail merge feature
12.	To use Equations and symbols features.	<p>a. Explore various symbols available in MS Word</p> <p>b. Insert a symbol in the text</p> <p>c. Insert mathematical equations in the document</p>	Enter Mathematical symbols and Equations in the word document
13.	To Practice with MS-EXCEL	<p>a. Open /create an MS Excel spread sheet and familiarize with MS Excel 2007 layout like MS office Button-</p> <p>b. Use Quick Access Toolbar- Title Bar- Ribbon- Worksheets- Formula Bar- Status Bar</p>	<p>a. Familiarize with excel layout and use</p> <p>b. Use various features available in toolbar</p>
14.	To access and Enter data in the cells	<p>a. Move Around a Worksheets- Quick access -Select Cells</p> <p>b. Enter Data-Edit a Cell-Wrap Text-Delete a Cell Entry-Save a File-Close Excel</p>	<p>a. Access and select the required cells by various addressing methods</p> <p>b. Enter data and edit</p>
15.	To edit spread sheet Copy, Cut, Paste, and selecting cells	<p>a. Insert and Delete Columns and Rows-Create Borders-Merge and Center</p> <p>b. Add Background Color-Change the Font, Font Size, and Font Color</p> <p>c. Format text with Bold, Italicize, and Underline-Work</p>	Format the excel sheet

		with Long Text-Change a Column's Width	
16.	To use built in functions and Formatting Data	<ul style="list-style-type: none"> <li>a. Perform Mathematical Calculations verify -AutoSum</li> <li>b. Perform Automatic Calculations-Align Cell Entries</li> </ul>	Use built in functions in Excel
17.	To enter a Formula for automatic calculations	<ul style="list-style-type: none"> <li>a. Enter formula</li> <li>b. Use Cell References in Formulae</li> <li>c. Use Automatic updating function of Excel Formulae</li> <li>d. Use Mathematical Operators in Formulae</li> <li>e. Use Excel Error Message and Help</li> </ul>	Enter formula for automatic calculations
18.	To Create Excel Functions, Filling Cells	<ul style="list-style-type: none"> <li>a. Use Reference Operators</li> <li>b. Work with sum, Sum if, Count and Count If Functions</li> <li>c. Fill Cells Automatically</li> </ul>	<ul style="list-style-type: none"> <li>a. Create Excel sheets involving cross references and equations</li> <li>b. Use the advanced functions for conditional calculations</li> </ul>
19.	To sort and filter data in table	<ul style="list-style-type: none"> <li>a. Sort data in multiple columns</li> <li>b. Sort data in a row</li> <li>c. Sort data using Custom order</li> <li>d. Filter data in work sheet</li> </ul>	<ul style="list-style-type: none"> <li>a. Refine the data in a worksheet and keep it organized</li> <li>b. Narrow a worksheet by selecting specific choice</li> </ul>
20.	To Practice Excel Graphs and Charts	<ul style="list-style-type: none"> <li>a. Produce an Excel Pie Chart</li> <li>b. Produce</li> <li>c. Excel Column Chart</li> </ul>	<ul style="list-style-type: none"> <li>a. Use data in Excel sheet to Create technical charts and graphs Produce Excel Line Graph</li> <li>b. Produce a Pictograph in Excel</li> </ul>
21.	To develop lab reports of respective discipline	Create Lab reports using MS Word and Excel	<ul style="list-style-type: none"> <li>a. Insert Practical subject name in Header and page numbers in Footer</li> </ul>
22.	To format a Worksheet in Excel, page setup and print	<ul style="list-style-type: none"> <li>a. Shade alternate rows of data</li> <li>b. Add currency and percentage symbols</li> </ul>	<ul style="list-style-type: none"> <li>a. Format Excel sheet</li> <li>b. Insert headers &amp; footers and print</li> </ul>

		<ul style="list-style-type: none"> <li>c. Change height of a row and width of a column</li> <li>d. Change data alignment</li> <li>e. Insert Headers and Footers</li> <li>f. Set Print Options and Print</li> </ul>	
23.	To familiarize with Ribbon layout & features of PowerPoint 2007.	<p>Use various options in PowerPoint</p> <ul style="list-style-type: none"> <li>a. Home</li> <li>b. Insert</li> <li>c. Design</li> <li>d. Animation</li> <li>e. Slideshow</li> <li>f. View</li> <li>g. Review</li> </ul>	Access required options in the tool bar
24.	To create a simple PowerPoint Presentation	<ul style="list-style-type: none"> <li>a. Insert a New Slide into PowerPoint</li> <li>b. Change the Title of a PowerPoint Slide</li> <li>c. PowerPoint Bullets</li> <li>d. Add an Image to a PowerPoint Slide</li> <li>e. Add a Textbox to a PowerPoint slide</li> </ul>	<ul style="list-style-type: none"> <li>a. Create simple PowerPoint presentation with photographs/ClipArt and text boxes</li> <li>b. Use bullets option</li> </ul>
25.	To Set up a Master Slide in PowerPoint and add notes	<ul style="list-style-type: none"> <li>a. Create a PowerPoint Design Template</li> <li>b. Modify themes</li> <li>c. Switch between Slide master view and Normal view</li> <li>d. Format a Design Template Master Slide</li> <li>e. Add a Title Slide to a Design Template</li> <li>f. The Slide Show Footer in PowerPoint</li> <li>f. Add Notes to a PowerPoint Presentation</li> </ul>	<ul style="list-style-type: none"> <li>a. Setup Master slide and format</li> <li>b. Add notes</li> </ul>
26.	To Insert Text and Objects	<ul style="list-style-type: none"> <li>a. Insert Text and objects</li> <li>b. Set Indents and line spacing</li> <li>c. Insert pictures/ clipart</li> <li>d. Format pictures</li> <li>e. Insert shapes and word art</li> <li>f. Use 3d features</li> <li>g. Arrange objects</li> </ul>	<p>Insert Text and Objects</p> <p>Use 3d features</p>

27.	To insert a Flow Chart / Organizational Charts	<ul style="list-style-type: none"> <li>a. Create a Flow Chart in PowerPoint</li> <li>b. Group and Ungroup Shapes</li> <li>c. Use smart art</li> </ul>	Create organizational charts and flow charts using smart art
28.	To insert a Table	<ul style="list-style-type: none"> <li>a. PowerPoint Tables</li> <li>b. Format the Table Data</li> <li>c. Change Table Background</li> <li>d. Format Series Legend</li> </ul>	Insert tables and format
29.	To insert a Charts/Graphs	<ul style="list-style-type: none"> <li>a. Create 3D Bar Graphs in PowerPoint</li> <li>b. Work with the PowerPoint Datasheet</li> <li>c. Format a PowerPoint Chart Axis</li> <li>d. Format the Bars of a Chart</li> <li>e. Create PowerPoint Pie Charts</li> <li>f. Use Pie Chart Segments</li> <li>g. Create 2D Bar Charts in PowerPoint</li> <li>h. Format the 2D Chart</li> <li>e. Format a Chart Background</li> </ul>	Create charts and Bar graphs, Pie Charts and format.
30.	To Insert audio & video, Hyperlinks in a slide Add narration to the slide	<ul style="list-style-type: none"> <li>a. Insert sounds in the slide and hide the audio symbol</li> <li>b. Adjust the volume in the settings</li> <li>c. Insert video file in the format supported by PowerPoint in a slide</li> <li>d. Use automatic and on click options</li> <li>e. Add narration to the slide</li> <li>f. Insert Hyperlinks</li> </ul>	<ul style="list-style-type: none"> <li>a. Insert Sounds and Video in appropriate format.</li> <li>b. Add narration to the slide</li> <li>c. Use hyperlinks to switch to different slides and files</li> </ul>
31.	To Practice Animation effects	<ul style="list-style-type: none"> <li>a. Apply transitions to slides</li> <li>b. To explore and practice special animation effects like <i>Entrance, Emphasis, Motion Paths &amp; Exit</i></li> </ul>	Add animation effects
32.	Reviewing presentation	<ul style="list-style-type: none"> <li>a. Checking spelling and grammar</li> <li>b. Previewing presentation</li> <li>c. Set up slide show</li> <li>d. Set up resolution</li> </ul>	<ul style="list-style-type: none"> <li>a. Use Spell check and Grammar feature</li> <li>b. Setup slide show</li> <li>c. Add timing to the slides</li> </ul>

		<ul style="list-style-type: none"> <li>e. Exercise with Rehearse Timings feature in PowerPoint</li> <li>f. Use PowerPoint Pen Tool during slide show</li> <li>g. Saving</li> <li>h. Printing presentation <ul style="list-style-type: none"> <li>(a) Slides</li> <li>(b) Hand-out</li> </ul> </li> </ul>	d. Setup automatic slide show
33	To familiarize with standard toolbox	<ul style="list-style-type: none"> <li>a. Open Adobe Photoshop</li> <li>b. Use various tools such as <ul style="list-style-type: none"> <li>i. The Layer Tool</li> <li>ii. The Color &amp; Swatches Tool</li> <li>iii. Custom Fonts &amp; The Text Tool</li> <li>iv. Brush Tool</li> <li>v. The Select Tool</li> <li>vi. The Move Tool</li> <li>vii. The Zoom Tool</li> <li>viii. The Eraser</li> <li>ix. The Crop Tool</li> <li>x. The Fill Tool</li> </ul> </li> </ul>	photograph and save it in Photoshop
34	To edit a photograph	<ul style="list-style-type: none"> <li>a. Use the Crop tool</li> <li>b. Trim edges</li> <li>c. Change the shape and size of a photo</li> <li>d. Remove the part of photograph including graphics and text</li> </ul>	to edit image by using corresponding tools.
35	To insert Borders around photograph	<ul style="list-style-type: none"> <li>a. Start with a single background layer</li> <li>b. Bring the background forward</li> <li>c. Enlarge the canvas</li> <li>d. Create a border color</li> <li>e. Send the border color to the back</li> <li>f. Experiment with different colors</li> </ul>	Able to create a border or frame around an image to add visual interest to a photo
36	To change Background of a Photograph	<ul style="list-style-type: none"> <li>a. open the foreground and background image</li> <li>b. Use different selection tools to paint over the image</li> <li>c. Copy background image and paste it on the foreground.</li> </ul>	Able to swap background elements using the Select and Mask tool and layers.

		<p>d. Resize and/or drag the background image to reposition.</p> <p>e. In the Layers panel, drag the background layer below the foreground image layer.</p>	
37	To change colors of Photograph	<p>a. Change colours using:</p> <p>i) Colour Replacement tool</p> <p>ii) Hue/Saturation adjustment layer tool</p>	Able to control color saturation
38	To prepare a cover page for the book in subject area	<p>a. open a file with height 500 and width 400 for the cover page.</p> <p>b. apply two different colours to work area by dividing it into two parts using Rectangle tool.</p> <p>c. Copy any picture and place it on work area → resize it using free transform tool.</p> <p>d. Type text and apply color and style</p> <p>e. Apply effects using blended options</p>	Able to prepare cover page for the book
39	To adjust the brightness and contrast of picture to give an elegant look	<p>a. open a file</p> <p>b. Go to image → adjustments → Brightness/Contrast.</p> <p>f. adjust the brightness and contrast</p> <p>g. save the image</p>	Able to control brightness/contrast.
40	To type a word and apply the shadow emboss effects	<p>a. open a file</p> <p>b. Select the text tool and type text.</p> <p>c. Select the typed text go to layer → layer style → blended option → drop shadow, inner shadow, bevel and emboss → contour → satin → gradient overlay</p> <p>d. Save the image.</p>	Able to apply shadow emboss effects

**Table specifying the scope of syllabus to be covered for unit tests**

<b>Unit Test</b>	<b>Learning outcomes to be covered</b>
Unit test-1	From 1 to 8
Unit test-2	From 9 to 22
Unit test-3	From 23 to 40